

NAMING IONIC COMPOUNDS

Name _____

Name the following compounds using the Stock Naming System.

and traditional
(where appropriate)

1. CaCO_3 calcium carbonate
2. KCl potassium chloride
3. $\overset{+2}{\text{Fe}}\text{SO}_4$ iron(II) sulfate ferrous sulfate
4. LiBr lithium bromide
5. MgCl_2 magnesium chloride
6. $\overset{+3}{\text{Fe}}\text{Cl}_3$ iron(III) chloride ferric chloride
7. $\text{Zn}_3(\text{PO}_4)_2$ zinc phosphate
8. NH_4NO_3 ammonium nitrate
9. $\text{Al}(\text{OH})_3$ aluminum hydroxide
10. $\overset{+1}{\text{Cu}}\text{C}_2\text{H}_3\text{O}_2$ copper(I) acetate cuprous acetate
11. $\overset{+2}{\text{Pb}}\text{SO}_3$ lead(II) sulfite plumbous sulfite
12. NaClO_3 sodium chlorate
13. CaC_2O_4 calcium oxalate
14. $\overset{+3}{\text{Fe}}_2\text{O}_3$ iron(III) oxide ferric oxide
15. $(\text{NH}_4)_3\text{PO}_4$ ammonium phosphate
16. NaHSO_4 sodium bisulfate / sodium hydrogen sulfate
17. Hg_2Cl_2 mercury(I) chloride mercurous chloride
18. $\text{Mg}(\text{NO}_2)_2$ magnesium nitrite
19. $\overset{+2}{\text{Cu}}\text{SO}_4$ copper(II) sulfate cupric sulfate
20. NaHCO_3 sodium bicarbonate / sodium hydrogen carbonate
21. $\overset{+3}{\text{Ni}}\text{Br}_2$ nickel(III) bromide
22. $\text{Be}(\text{NO}_3)_2$ beryllium nitrate
23. ZnSO_4 zinc sulfate
24. $\overset{+3}{\text{Au}}\text{Cl}_3$ gold(III) chloride
25. KMnO_4 potassium permanganate

Naming Compounds Containing Polyatomic Ions

Answers

M Tight



Naming Compounds Containing Polyatomic Ions Answers:

The Practice of Chemistry Donald J. Wink, Sharon Fetzner-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math. The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving. The book's unique focus on actual chemical practice, extensive study tools, and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors. This special PowerPoint tour of the text was created by Don Wink. http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt 832KB

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Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts. After a brief review of each section's critical ideas, students are taken through stepped-out worked examples, try-it-yourself examples, and chapter quizzes, all structured to reinforce chapter objectives and build problem-solving techniques. The solutions manual includes detailed solutions to all odd-numbered exercises in the text.

Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry: The Molecular Nature of Matter, 8th Edition, continues to focus on the intimate relationship that exists between structure at the atomic/molecular level and the observable macroscopic properties of matter. Key revisions in this edition focus on three areas: The deliberate inclusion of more updated real-world examples that relate common real-world student experiences to the science of chemistry. Simultaneously, examples and questions have been updated to align them with career concepts relevant to the environmental

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Regents Chemistry--Physical Setting Power Pack Revised Edition Barron's Educational Series, Albert S. Tarendash, 2021-01-05 Barron's two book Regents Chemistry Power Pack provides comprehensive review actual administered exams and practice questions to help students prepare for the Chemistry Regents exam This edition includes Regents Exams and Answers Chemistry Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day Let's Review Regents Chemistry Extensive review of all topics on the test Extra practice questions with answers A detailed introduction to the Regents Chemistry course and exam One actual recently released Regents Chemistry exam with an answer key

Introduction to General, Organic, and Biochemistry Morris Hein, Scott Pattison, Susan Arena, Leo R. Best, 2014-01-15 The most comprehensive book available on the subject Introduction to General Organic and Biochemistry 11th Edition continues its tradition of fostering the development of problem solving skills featuring numerous examples and coverage of current applications Skillfully anticipating areas of difficulty and pacing the material accordingly this readable work provides clear and logical explanations of chemical concepts as well as the right mix of general chemistry organic chemistry and biochemistry An emphasis on real world topics lets readers clearly see how the chemistry will apply to their career

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Foundations of College Chemistry Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded

coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van t Hoff factor and applications of radioactivity Survival Guide to General Chemistry Patrick E.

McMahon,Rosemary McMahon,Bohdan Khomtchouk,2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields **Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual** Leo J. Malone,Theodore O.

Dolter,2012-01-03 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter Let's Review Regents:

Chemistry--Physical Setting Revised Edition Barron's Educational Series,Albert S. Tarendash,2021-01-05 Barron s Let s Review Regents Chemistry gives students the step by step review and practice they need to prepare for the Regents Chemistry Physical Setting exam This updated edition is an ideal companion to high school textbooks and covers all Chemistry topics prescribed by the New York State Board of Regents Let s Review Regents Chemistry covers all high school level Chemistry topics and includes Extensive review of all topics on the test Extra practice questions with answers A

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Jones,Leroy Laverman,2016-01-07 Written for calculus inclusive general chemistry courses Chemical Principles helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry s frontier continually demonstrating how to solve problems think about nature and matter and visualize chemical concepts as working chemists do It also offers an exceptional level of support to help students develop their mathematical and problem solving skills For the new edition Chemical Principles now takes a modular approach with coverage organized as a series of brief Topics within 13 major areas of focus including a refresher on the fundamentals of chemistry and an online only section on techniques *E3 Chemistry Review Book - 2018 Home Edition (Answer Key Included)* Effiong Eyo,2017-10-20 With Answer Key to All Questions Chemistry students and homeschoolers Go beyond just passing Enhance your understanding of chemistry and get higher marks on homework quizzes tests and the regents exam with E3 Chemistry Review Book 2018 With E3 Chemistry Review Book students will get clean clear engaging exciting and easy to understand high school chemistry concepts with

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the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

An Introduction to Chemistry Michael Mosher, Paul Kelter, 2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life It provides a major overview of topics typically found in first year chemistry courses in the USA The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry

Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *Lab Manual for General, Organic, and Biochemistry* Denise Guinn, Rebecca Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives *Essentials of General Organic and Biochemistry* captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession To browse a sample chapter view sample ChemCasts and more visit www.whfreeman.com/gob

Unveiling the Magic of Words: A Review of "**Naming Compounds Containing Polyatomic Ions Answers**"

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