

Assignment #2 – Compound Names and Formulas
Multi-valent ions only

A. Write the correct formula for the following compounds, all of which have been named using the modern Roman Numeral Method. The ionic charge is given after the first element (metallic) in Roman Numerals.

1. copper (II) oxide CuO
2. mercury (I) oxide Hg₂O
3. gold (III) chloride AuCl₃
4. thallium (III) bromide TlBr₃
5. bismuth (V) oxide Bi₂O₅
6. terbium (III) oxide Tb₂O₃
7. uranium (VI) oxide UO₃
8. protactinium (IV) bromide PaBr₄
9. cerium (III) oxide Ce₂O₃
10. arsenic (V) sulphide As₂S₅

11. manganese (III) oxide Mn₂O₃
12. vanadium (II) bromide VBr₂
13. niobium(V) oxide Nb₂O₅
14. titanium (III) oxide Ti₂O₃
15. titanium (III) nitride TiN
16. iron (II) oxide FeO
17. cobalt (II) phosphide Co₃P₂
18. tin (II) oxide SnO
19. thulium (II) bromide TmBr₂
20. copper (II) bromide CuBr₂

B. Name the compound using the Roman Numeral Method and balance the ionic charges.

1. SnCl₄ Tin (IV) Chloride
2. BiBr₅ Bismuth (V) Bromide
3. PoO₂ Polonium (IV) Oxide
4. PbI₂ Lead (II) Iodide
5. HgO Mercury (II) Oxide
6. HgCl Mercury (I) Chloride
7. Au₂O₃ Gold (III) Oxide
8. FeCl₂ Iron (II) Chloride

9. PdF₄ Palladium (IV) Fluoride
10. Os₂O₃ Osmium (III) Oxide
11. MoBr₂ Molybdenum (II) Bromide
12. VCl₅ Vanadium (V) Chloride
13. Mn₂O₃ Manganese (III) Oxide
14. CoO Cobalt (II) Oxide
15. Np₂O₃ Neptunium (III) Oxide
16. V₂O₅ Vanadium (V) Oxide

Naming Balancing And Formula Writing Answer Key

Lingjun Ying



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Table of Contents Naming Balancing And Formula Writing Answer Key

1. Understanding the eBook Naming Balancing And Formula Writing Answer Key
 - The Rise of Digital Reading Naming Balancing And Formula Writing Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Naming Balancing And Formula Writing Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Naming Balancing And Formula Writing Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Naming Balancing And Formula Writing Answer Key
 - Personalized Recommendations
 - Naming Balancing And Formula Writing Answer Key User Reviews and Ratings
 - Naming Balancing And Formula Writing Answer Key and Bestseller Lists
5. Accessing Naming Balancing And Formula Writing Answer Key Free and Paid eBooks
 - Naming Balancing And Formula Writing Answer Key Public Domain eBooks
 - Naming Balancing And Formula Writing Answer Key eBook Subscription Services
 - Naming Balancing And Formula Writing Answer Key Budget-Friendly Options

6. Navigating Naming Balancing And Formula Writing Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Naming Balancing And Formula Writing Answer Key Compatibility with Devices
 - Naming Balancing And Formula Writing Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Naming Balancing And Formula Writing Answer Key
 - Highlighting and Note-Taking Naming Balancing And Formula Writing Answer Key
 - Interactive Elements Naming Balancing And Formula Writing Answer Key
8. Staying Engaged with Naming Balancing And Formula Writing Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Naming Balancing And Formula Writing Answer Key
9. Balancing eBooks and Physical Books Naming Balancing And Formula Writing Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Naming Balancing And Formula Writing Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Naming Balancing And Formula Writing Answer Key
 - Setting Reading Goals Naming Balancing And Formula Writing Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Naming Balancing And Formula Writing Answer Key
 - Fact-Checking eBook Content of Naming Balancing And Formula Writing Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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