

11. A compound X, of boron reacts with NH_3 on heating to give another compound Y which is called inorganic benzene. The compound X can be prepared by treating BF_3 with Lithium aluminium hydride. The compounds X and Y are represented by the formulas.
- (i) B_2H_6 , $\text{B}_3\text{N}_3\text{H}_6$
 - (ii) B_2O_3 , $\text{B}_3\text{N}_3\text{H}_6$
 - (iii) BF_3 , $\text{B}_3\text{N}_3\text{H}_6$
 - (iv) $\text{B}_3\text{N}_3\text{H}_6$, B_2H_6
12. Quartz is extensively used as a piezoelectric material, it contains _____.
- (i) Pb
 - (ii) Si
 - (iii) Ti
 - (iv) Sn
13. The most commonly used reducing agent is
- (i) AlCl_3
 - (ii) PbCl_2
 - (iii) SnCl_4
 - (iv) SnCl_2
14. Dry ice is
- (i) Solid NH_3
 - (ii) Solid SO_2
 - (iii) Solid CO_2
 - (iv) Solid N_2
15. Cement, the important building material is a mixture of oxides of several elements. Besides calcium, iron and sulphur, oxides of elements of which of the group (s) are present in the mixture?
- (i) group 2
 - (ii) groups 2, 13 and 14
 - (iii) groups 2 and 13
 - (iv) groups 2 and 14

II. Multiple Choice Questions (Type-II)

In the following questions two or more options may be correct.

16. The reason for small radius of Ga compared to Al is _____.
- (i) poor screening effect of d and f orbitals
 - (ii) increase in nuclear charge

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