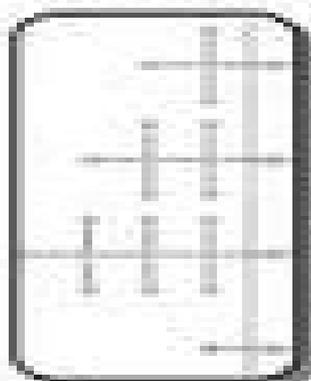


## Naming Substituted Hydrocarbons

A substituted hydrocarbon is a hydrocarbon with an atom or atoms other than hydrogen attached to carbon atoms along the hydrocarbon chain. It is usually a smaller hydrocarbon or a hydrocarbon. This can be illustrated with alcohols as an example. The compounds produced in the lower left are alcohols. They look like alkanes with  $-OH$  at one end where a hydrogen would have been. The  $-OH$  is called a functional group. The one with alcohols is called a primary alcohol (1<sup>o</sup>). The general formula for alcohols is  $C_nH_{2n+2}O$ . The first alcohol produced in the list is named by removing an  $-H$  from the hydrocarbon molecule  $C_2H_6$ . An alcohol is called a primary alcohol. The name indicates that it is an alcohol. The next carbon atom from the carbon is called the number 1 carbon and the rest of the  $-OH$ . The next alcohol in the series,  $C_3H_8O$ , formed from ethane, is  $CH_3CH_2CH_2OH$ .

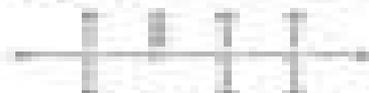


$CH_3CH_2CH_2CH_3$  is called 1-propanol.

The alcohols and other organic classes of substituted hydrocarbons are listed in Table 10. The names determined by counting the number of carbons in the chain. For alcohols, the substituents are identified with a prefix. For the remaining hydrocarbons, a prefix is used. From Table 10, the name of hydrocarbons, the number and location of groups needs to be identified.



1,2,2-trimethylpropane



2-butanol

Name the following compounds using the rules for naming hydrocarbons and by referring to the ending and Table 10 above.

1.  $CH_3CH_2CH_2CH_2CH_2CH_3$  \_\_\_\_\_



2. \_\_\_\_\_

Table 10  
Organic Functional Groups

Class of Compound	Functional Group	General Formula	Example
Alkyl Hydrocarbon	$-CH_3$ Methyl $-C_2H_5$ Ethyl $-C_3H_7$ Propyl $-C_4H_9$ Butyl	$C_nH_{2n+2}$	1-propanol, 2-butanol, 2-propanol
Alcohol	$-OH$	$C_nH_{2n+2}O$	1-propanol, 2-butanol, 2-propanol
Aldehyde	$-CHO$	$C_nH_{2n}O$	1-propanol, 2-butanol, 2-propanol
Ketone	$-CO-$	$C_nH_{2n}O$	1-propanol, 2-butanol, 2-propanol
Carboxylic acid	$-COOH$	$C_nH_{2n}O_2$	1-propanol, 2-butanol, 2-propanol
Amine	$-NH_2$	$C_nH_{2n+3}N$	1-propanol, 2-butanol, 2-propanol
Nitrile	$-CN$	$C_nH_{2n-1}N$	1-propanol, 2-butanol, 2-propanol
Halide	$-X$	$C_nH_{2n+2}X$	1-propanol, 2-butanol, 2-propanol

# Naming Substituted Hydrocarbons Answers

**David E. Henderson, Susan K.  
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