

## **Naming Ionic Compounds – Answer Key**

*Give the name of the following ionic compounds:*

		<b>Name</b>
1)	$\text{Na}_2\text{CO}_3$	<b>sodium carbonate</b>
2)	$\text{NaOH}$	<b>sodium hydroxide</b>
3)	$\text{MgBr}_2$	<b>magnesium bromide</b>
4)	$\text{KCl}$	<b>potassium chloride</b>
5)	$\text{FeCl}_2$	<b>iron (II) chloride</b>
6)	$\text{FeCl}_3$	<b>iron (III) chloride</b>
7)	$\text{Zn(OH)}_2$	<b>zinc hydroxide</b>
8)	$\text{Be}_2\text{SO}_4$	<b>beryllium sulfate</b>
9)	$\text{CrF}_2$	<b>chromium (II) fluoride</b>
10)	$\text{Al}_2\text{S}_3$	<b>aluminum sulfide</b>
11)	$\text{PbO}$	<b>lead (II) oxide</b>
12)	$\text{Li}_3\text{PO}_4$	<b>lithium phosphate</b>
13)	$\text{TiI}_4$	<b>titanium (IV) iodide</b>
14)	$\text{Co}_3\text{N}_2$	<b>cobalt (II) nitride</b>
15)	$\text{Mg}_3\text{P}_2$	<b>magnesium phosphide</b>
16)	$\text{Ga(NO}_2)_3$	<b>gallium nitrite</b>
17)	$\text{Ag}_2\text{SO}_3$	<b>silver sulfite</b>
18)	$\text{NH}_4\text{OH}$	<b>ammonium hydroxide</b>
19)	$\text{Al(CN)}_3$	<b>aluminum cyanide</b>
20)	$\text{Be(CH}_3\text{COO)}_2$	<b>beryllium acetate</b>

# Naming Ionic Compounds Key

**N Noddings**



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