

### Assignment #3 – Compound Names and Formulas Single and Multi-valent ions

A. Write the correct formula for the following compounds.

1. gallium oxide  $Ga_2O_3$
2. mercury (I) chloride  $HgCl$
3. tin (IV) bromide  $SnBr_4$
4. indium oxide  $In_2O_3$
5. cadmium sulphide  $CdS$
6. magnesium oxide  $MgO$
7. potassium nitride  $K_3N$
8. iron (III) sulphide  $Fe_2S_3$
9. gold (III) bromide  $AuBr_3$
10. zinc oxide  $ZnO$

11. silicon oxide  $SiO_2$
12. aluminum bromide  $AlBr_3$
13. tin (II) sulphide  $SnS$
14. germanium bromide  $GeBr_4$
15. nickel (II) oxide  $NiO$
16. strontium astatide  $SrAt_2$
17. aluminum oxide  $Al_2O_3$
18. copper (II) oxide  $CuO$
19. lawrencium fluoride  $LrF_3$
20. actinium oxide  $Ac_2O_3$

B. Name the following compounds correctly. Only include the Roman Numeral in the name if the first element has 2 or more combining capacities.

1.  $CaBr_2$  Calcium Bromide
2.  $K_2O$  Potassium Oxide
3.  $Fe_2O_3$  Iron (III) Oxide
4.  $ZnF_2$  Zinc Fluoride
5.  $CuCl$  Copper (I) Chloride
6.  $Ni_2O_3$  Nickel (III) Oxide
7.  $AuCl$  Gold (I) Chloride
8.  $HgBr$  Mercury (I) Bromide

9.  $AsCl_5$  Arsenic (V) Chloride
10.  $Eu_2O_3$  Europium (III) Oxide
11.  $NdBr_3$  Neodymium Bromide
12.  $Cu_3N_2$  Copper (II) Nitride
13.  $CoS$  Cobalt (II) Sulphide
14.  $ZrO_2$  Zirconium Oxide
15.  $Li_2O$  Lithium Oxide
16.  $NaH$  Sodium Hydride

# Naming Ionic Compounds If8766 Answer

**RJ Shavelson**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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### **Table of Contents Naming Ionic Compounds If8766 Answer**

1. Understanding the eBook Naming Ionic Compounds If8766 Answer
  - The Rise of Digital Reading Naming Ionic Compounds If8766 Answer
  - Advantages of eBooks Over Traditional Books
2. Identifying Naming Ionic Compounds If8766 Answer
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Naming Ionic Compounds If8766 Answer
  - User-Friendly Interface
4. Exploring eBook Recommendations from Naming Ionic Compounds If8766 Answer

- Personalized Recommendations
- Naming Ionic Compounds If8766 Answer User Reviews and Ratings
- Naming Ionic Compounds If8766 Answer and Bestseller Lists
- 5. Accessing Naming Ionic Compounds If8766 Answer Free and Paid eBooks
  - Naming Ionic Compounds If8766 Answer Public Domain eBooks
  - Naming Ionic Compounds If8766 Answer eBook Subscription Services
  - Naming Ionic Compounds If8766 Answer Budget-Friendly Options
- 6. Navigating Naming Ionic Compounds If8766 Answer eBook Formats
  - ePub, PDF, MOBI, and More
  - Naming Ionic Compounds If8766 Answer Compatibility with Devices
  - Naming Ionic Compounds If8766 Answer Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Naming Ionic Compounds If8766 Answer
  - Highlighting and Note-Taking Naming Ionic Compounds If8766 Answer
  - Interactive Elements Naming Ionic Compounds If8766 Answer
- 8. Staying Engaged with Naming Ionic Compounds If8766 Answer
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Naming Ionic Compounds If8766 Answer
- 9. Balancing eBooks and Physical Books Naming Ionic Compounds If8766 Answer
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Naming Ionic Compounds If8766 Answer
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Naming Ionic Compounds If8766 Answer
  - Setting Reading Goals Naming Ionic Compounds If8766 Answer
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Naming Ionic Compounds If8766 Answer

- Fact-Checking eBook Content of Naming Ionic Compounds If8766 Answer
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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