



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
(Show the steps for naming the compound)



Steps for Naming:

1. Identify the cation (metal) and anion (non-metal).
2. Write the name of the cation first, followed by the name of the anion.
3. Use the appropriate prefix for the number of atoms of each element.
4. The name of the anion ends with the suffix -ide.
5. The name of the compound is Carbon Dioxide.

Prefixes

Prefix	Number
mono-	1
di-	2
tri-	3
tetra-	4
penta-	5
hexa-	6
hepta-	7
octa-	8
nona-	9
deca-	10

Common Ions

Ion	Name
H ⁺	Hydrogen
OH ⁻	Hydroxide
Cl ⁻	Chloride
Br ⁻	Bromide
I ⁻	Iodide
S ²⁻	Sulfide
SO ₄ ²⁻	Sulfate
NO ₃ ⁻	Nitrate
CO ₃ ²⁻	Carbonate

Naming Compounds with a Chemical Formula



Carbon Dioxide

Steps for Naming:

1. Identify the cation (metal) and anion (non-metal).
2. Write the name of the cation first, followed by the name of the anion.
3. Use the appropriate prefix for the number of atoms of each element.
4. The name of the anion ends with the suffix -ide.
5. The name of the compound is Carbon Dioxide.

Prefixes

Prefix	Number	Prefix	Number
mono-	1	deca-	10
di-	2	hepta-	7
tri-	3	hexa-	6
tetra-	4	penta-	5
penta-	5	octa-	8
hexa-	6	nona-	9
hepta-	7	deca-	10

Common Ions for Specific Compounds

Ion	Name	Ion	Name
H ⁺	Hydrogen	OH ⁻	Hydroxide
Cl ⁻	Chloride	SO ₄ ²⁻	Sulfate
NO ₃ ⁻	Nitrate	CO ₃ ²⁻	Carbonate
Br ⁻	Bromide	PO ₄ ³⁻	Phosphate

Naming Compounds Study Guide

Daniela Niemeyer



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