

# NAMING MOLECULAR COMPOUNDS

Name \_\_\_\_\_

Name the following covalent compounds.

1.  $\text{CO}_2$  Carbon dioxide
2.  $\text{CO}$  Carbon monoxide
3.  $\text{SO}_2$  Sulfur dioxide
4.  $\text{SO}_3$  Sulfur trioxide
5.  $\text{N}_2\text{O}$  Dinitrogen oxide
6.  $\text{NO}$  nitrogen oxide
7.  $\text{N}_2\text{O}_3$  Dinitrogen trioxide
8.  $\text{NO}_2$  Nitrogen dioxide
9.  $\text{N}_2\text{O}_4$  Dinitrogen tetroxide
10.  $\text{N}_2\text{O}_5$  Dinitrogen Pentoxide
11.  $\text{PCl}_3$  Phosphorus trichloride
12.  $\text{PCl}_5$  Phosphorus pentachloride
13.  $\text{NH}_3$  Nitrogen trihydride
14.  $\text{SCl}_6$  Sulfur hexachloride
15.  $\text{P}_2\text{O}_5$  Diphosphorus pentoxide
16.  $\text{CCl}_4$  Carbon tetrachloride
17.  $\text{SiO}_2$  Silicon dioxide
18.  $\text{CS}_2$  Carbon disulfide
19.  $\text{OF}_2$  oxygen difluoride
20.  $\text{PBr}_3$  Phosphorus tribromide

- 1- mono
- 2- di
- 3- tri
- 4- tetra
- 5- penta
- 6- hexa

# Naming Binary Covalent Compounds Answer Key

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