

NACE CIP Level 1 Exam Study Guide With 100% Complete Solution 2024

Define Corrosion - Correct Answer-The deterioration of a substance, usually a metal from a reaction with its environment

What is passivation? - Correct Answer-A layer of oxides formed on the surface of a metal that provides corrosion protection.

What is an example of passivation? - Correct Answer-Stainless Steel

What are the elements of a corrosion cell? - Correct Answer-Anode, Cathode, Metallic Pathway, Electrolyte

Describe what happens at the anode. - Correct Answer-The metal dissolves into the electrolyte

What is the function of the electrolyte? - Correct Answer-Allows the passage of ions to connect the corrosion cell

What is the function of the metallic pathway? - Correct Answer-Allows the flow of electrons to connect the corrosion cell

What is the galvanic series? - Correct Answer-A listing of metals in order of reactivity (more or less noble) in seawater at 25 C

The general rules of galvanic corrosion are: - Correct Answer-The less noble (or more reactive) metal when connected to a more noble (or less reactive metal) will corrode preferentially

Name the five most important factors that affect the rate of corrosion. - Correct Answer-Humidity, oxygen, chemical salts, pollutants, temperature

General corrosion is: - Correct Answer-Results in a relatively uniform loss of material over the entire surface, Results in a general thinning of the affected surface, Relatively easy to inspect, Does not cause catastrophic failures

Localized corrosion is: - Correct Answer-Typically pitting and crevice corrosion. It is typically of the most concern as damage is concentrated and loss of integrity or structural failure can result.

List some of the common tools/methods used for corrosion control. - Correct Answer-Design, Inhibitors, Material Selection, Cathodic Protection, Protective Coatings, Splash Zone Systems, Alteration of the Environment

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Lingsheng Yao



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