

Balancing Equations Worksheet

A. Write the correct coefficients to balance each equation below

- 1) _____ Na_3PO_4 + _____ KOH \rightarrow _____ NaOH + _____ K_3PO_4
- 2) _____ MgF_2 + _____ Li_2CO_3 \rightarrow _____ MgCO_3 + _____ LiF
- 3) _____ P_4 + _____ O_2 \rightarrow _____ P_2O_5
- 4) _____ RbNO_3 + _____ BeF_2 \rightarrow _____ $\text{Be}(\text{NO}_3)_2$ + _____ RbF
- 5) _____ AgNO_3 + _____ Cu \rightarrow _____ $\text{Cu}(\text{NO}_3)_2$ + _____ Ag
- 6) _____ CF_4 + _____ Br_2 \rightarrow _____ CBr_4 + _____ F_2
- 7) _____ HCN + _____ CuSO_4 \rightarrow _____ H_2SO_4 + _____ $\text{Cu}(\text{CN})_2$
- 8) _____ GaF_3 + _____ Cs \rightarrow _____ CsF + _____ Ga
- 9) _____ BaS + _____ PtF_2 \rightarrow _____ BaF_2 + _____ PtS
- 10) _____ N_2 + _____ H_2 \rightarrow _____ NH_3
- 11) _____ NaF + _____ Br_2 \rightarrow _____ NaBr + _____ F_2
- 12) _____ $\text{Pb}(\text{OH})_2$ + _____ HCl \rightarrow _____ H_2O + _____ PbCl_2
- 13) _____ AlBr_3 + _____ K_2SO_4 \rightarrow _____ KBr + _____ $\text{Al}_2(\text{SO}_4)_3$
- 14) _____ CH_4 + _____ O_2 \rightarrow _____ CO_2 + _____ H_2O
- 15) _____ Na_3PO_4 + _____ CaCl_2 \rightarrow _____ NaCl + _____ $\text{Ca}_3(\text{PO}_4)_2$
- 16) _____ K + _____ Cl_2 \rightarrow _____ KCl
- 17) _____ Al + _____ HCl \rightarrow _____ H_2 + _____ AlCl_3
- 18) _____ N_2 + _____ F_2 \rightarrow _____ NF_3
- 19) _____ SO_2 + _____ Li_2Se \rightarrow _____ SSe_2 + _____ Li_2O
- 20) _____ NH_3 + _____ H_2SO_4 \rightarrow _____ $(\text{NH}_4)_2\text{SO}_4$

B. Complete the following tasks-

1. For any two equations above, find out by researching if the reaction requires energy or produces energy.
2. For any two equations above, find out the common chemical names for all reactants and products.
3. Find and describe any other chemical equation that involves water.

Equation Writing And Predicting Products Lab Answer Key

Richard Bailey



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