

Unit 2 Study Guide: Matter and Energy

Section 3.1 Properties of Matter

Use each of the terms below just once to complete the passage.

chemical
density

mass
properties

physical
substance

Matter is anything with (1) _____ and volume. A (2) _____ is a form of matter with a uniform and unchanging composition. Substances have specific, unchanging (3) _____ that can be observed. Substances have both physical and chemical properties. (4) _____ properties can be observed without changing a substance's chemical composition. Color, hardness, and (5) _____ are examples. Other properties cannot be observed without changing the composition of a substance. These are called (6) _____ properties. An example is the tendency of iron to form rust when exposed to air.

Label each property as either physical or chemical.

- _____ 7. Chemical formula is H_2O
- _____ 8. Forms green carbonate when exposed to moist air
- _____ 9. Remains unchanged when in the presence of nitrogen
- _____ 10. Colorless
- _____ 11. Solid at normal temperatures and pressures
- _____ 12. Ability to combine with another substance
- _____ 13. Melting point
- _____ 14. Liquid at normal temperatures and pressures
- _____ 15. Boiling point is 100°C
- _____ 16. Conducts electricity
- _____ 17. Density is $\frac{1\text{g}}{\text{cm}^3}$

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Table of Contents Energy Study Guide Typepad

1. Understanding the eBook Energy Study Guide Typepad
 - The Rise of Digital Reading Energy Study Guide Typepad
 - Advantages of eBooks Over Traditional Books
2. Identifying Energy Study Guide Typepad
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Energy Study Guide Typepad
 - User-Friendly Interface
4. Exploring eBook Recommendations from Energy Study Guide Typepad
 - Personalized Recommendations
 - Energy Study Guide Typepad User Reviews and Ratings
 - Energy Study Guide Typepad and Bestseller Lists
5. Accessing Energy Study Guide Typepad Free and Paid eBooks

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- Energy Study Guide Typepad Budget-Friendly Options
- 6. Navigating Energy Study Guide Typepad eBook Formats
 - ePub, PDF, MOBI, and More
 - Energy Study Guide Typepad Compatibility with Devices
 - Energy Study Guide Typepad Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Energy Study Guide Typepad
 - Highlighting and Note-Taking Energy Study Guide Typepad
 - Interactive Elements Energy Study Guide Typepad
- 8. Staying Engaged with Energy Study Guide Typepad
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Energy Study Guide Typepad
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 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Energy Study Guide Typepad
 - Setting Reading Goals Energy Study Guide Typepad
 - Carving Out Dedicated Reading Time
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 - Fact-Checking eBook Content of Energy Study Guide Typepad
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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