

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- | | |
|---------------|---|
| <u>yes</u> | 1. A tomato smells rotten. |
| <u>no</u> | 2. A drinking glass breaks into smaller pieces. |
| <u>no</u> | 3. A piece of ice melts. |
| <u>yes</u> | 4. Drain cleaner is mixed with water and the solution becomes warm. |
| <u>yes</u> | 5. Candle wax burns. |
| <u>yes/no</u> | 6. Molten candle wax solidifies. |
| <u>yes</u> | 7. Green leaves turn yellow and red as the seasons change. |
| <u>yes</u> | 8. Baking powder produces a gas that makes a cake rise. |

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** reactant, oxygen is a(n) **(10)** reactant, and water vapor is a(n) **(11)** product. In a chemical equation for this reaction, a(n) **(12)** arrow is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** plus sign is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** (g) symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** (g) symbol is used for the state of oxygen, and a(n) **(16)** (g) symbol is used for the state of water vapor.

Energy And Chemical Reactions Study Questions Answers

CL Gary



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Table of Contents Energy And Chemical Reactions Study Questions Answers

1. Understanding the eBook Energy And Chemical Reactions Study Questions Answers
 - The Rise of Digital Reading Energy And Chemical Reactions Study Questions Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Energy And Chemical Reactions Study Questions Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Energy And Chemical Reactions Study Questions Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Energy And Chemical Reactions Study Questions Answers
 - Personalized Recommendations
 - Energy And Chemical Reactions Study Questions Answers User Reviews and Ratings
 - Energy And Chemical Reactions Study Questions Answers and Bestseller Lists

5. Accessing Energy And Chemical Reactions Study Questions Answers Free and Paid eBooks
 - Energy And Chemical Reactions Study Questions Answers Public Domain eBooks
 - Energy And Chemical Reactions Study Questions Answers eBook Subscription Services
 - Energy And Chemical Reactions Study Questions Answers Budget-Friendly Options
6. Navigating Energy And Chemical Reactions Study Questions Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Energy And Chemical Reactions Study Questions Answers Compatibility with Devices
 - Energy And Chemical Reactions Study Questions Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Energy And Chemical Reactions Study Questions Answers
 - Highlighting and Note-Taking Energy And Chemical Reactions Study Questions Answers
 - Interactive Elements Energy And Chemical Reactions Study Questions Answers
8. Staying Engaged with Energy And Chemical Reactions Study Questions Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Energy And Chemical Reactions Study Questions Answers
9. Balancing eBooks and Physical Books Energy And Chemical Reactions Study Questions Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Energy And Chemical Reactions Study Questions Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Energy And Chemical Reactions Study Questions Answers
 - Setting Reading Goals Energy And Chemical Reactions Study Questions Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Energy And Chemical Reactions Study Questions Answers
 - Fact-Checking eBook Content of Energy And Chemical Reactions Study Questions Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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