#### Worksheet: Calculating Empirical & Molecular Formulas

1. The empirical formula for the compound having the formula H<sub>2</sub>C<sub>2</sub>O<sub>3</sub> is:

ISS CO'H SCI COH [D] CyOuHe

Calculate the empirical formula of a compound that is 85.6% C and 14.4% H (by mass).

BALCH, 181 CH TOT CVH-IDI C.H. (E) C.H.

 A compound is analyzed and found to contain 12.1% carbon, 16.2% oxygen, and 71.7% chlorine (by mass). Calculate the empirical formula of this compound.

C (2.1 - 1 1 U 767 = 2 ( [All COCI, ] [B] CO<sub>2</sub>CI, (C) CO2CI IDI COCH [E] COCK O Thomas

4. A compound contains 40.0% carbon, 6.7% hydrogen, and 53.3% oxygen (by mass). Colculate the empirical formula.

74.04 g 0

 Determine the empirical formula of a compound containing 54.2% F and 45.8% S (by mass).

19.00 g/med = 2.85 med F = 1,99 ~ 2 32-079/mol = 1.43 mol S = 1

T. A compound has 40.68% carbon, 5.12% hydrogen, and 54.20% oxygen (by mass). Calculate its empirical formula.

40.683C = 3.387 5.129 H = 5.669 54.209 = 3.387 = 1.496 x2= 3 = 1 ×2 = 2

8. A 7.33-g sample of lanthanum, La, combines with oxygen to give 10.29 g of the oxide. Calculate the empirical formula of this oxide.

10.29g - 7.33g La = 2.96g & 7.339 La = .0027 2.9490 = .1850 = 3.5 x2 = 7

9. Calculate the molecular formula of a compound with the empirical formula CH<sub>2</sub>O and a moler mass. of 150 gimel. 150 - 30 = 5 200

[A] C,H,O, [B] C,H,O, [C] C,H,O, EDIT CLINLOW DELICATION OF

# **Empirical Formula Study Guide With Answer Sheet**

Scott C. Dulebohn

**Empirical Formula Study Guide With Answer Sheet:** 

The book delves into Empirical Formula Study Guide With Answer Sheet. Empirical Formula Study Guide With Answer Sheet is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Empirical Formula Study Guide With Answer Sheet, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
  - Chapter 1: Introduction to Empirical Formula Study Guide With Answer Sheet
  - o Chapter 2: Essential Elements of Empirical Formula Study Guide With Answer Sheet
  - o Chapter 3: Empirical Formula Study Guide With Answer Sheet in Everyday Life
  - Chapter 4: Empirical Formula Study Guide With Answer Sheet in Specific Contexts
  - $\circ$  Chapter 5: Conclusion
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- 3. In chapter 2, the author will delve into the foundational concepts of Empirical Formula Study Guide With Answer Sheet. The second chapter will elucidate the essential principles that need to be understood to grasp Empirical Formula Study Guide With Answer Sheet in its entirety.
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- 6. In chapter 5, the author will draw a conclusion about Empirical Formula Study Guide With Answer Sheet. The final chapter will summarize the key points that have been discussed throughout the book.

  This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly
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