

Empirical and Molecular Formula Practice KEY

Empirical Formula

1. 36.5% Na, 25.4% S, and 38.1% O forms this compound.

$$\text{Na} - 36.5/23 = 1.56/0.8 = 2$$

$$\text{S} - 25.4/32 = 0.8/0.8 = 1$$

$$\text{O} - 38.1/16 = 2.4/0.8 = 3$$



2. A compound made of 72.2% Magnesium and 27.8% nitrogen by mass.

$$\text{Mg} - 72.2/24.3 = 3/2 = 1.5(2) = 3$$

$$\text{N} - 27.8/14 = 2/2 = 1(2) = 2$$



3. A chromium compound with 68.4% Cr and 31.6% O.

$$\text{Cr} - 68.4/52 = 1.315/1.315 = 1(2) = 2$$

$$\text{O} - 31.6/16 = 1.975/1.315 = 1.5(2) = 3$$



4. A compound with the composition of 24% C, 7% H, 38% F, and 31% P.

$$\text{C} - 24/12 = 2/1 = 2$$

$$\text{H} - 7/1 = 7/1 = 7$$

$$\text{F} - 38/19 = 2/1 = 2$$

$$\text{P} - 31/31 = 1/1 = 1$$



5. 63.5% silver, 8.2% nitrogen, and 28.3% oxygen make up this compound.

$$\text{Ag} - 63.5/107.87 = 0.589/0.585 = 1$$

$$\text{N} - 8.2/14.01 = 0.585/0.585 = 1$$

$$\text{O} - 28.3/16 = 1.769/0.585 = 3$$



6. 8.81g C and 91.2g Chlorine forms this compound.

$$\text{C} - 8.81/12 = 0.73/0.73 = 1(2) = 2$$

$$\text{Cl} - 91.2/35 = 2.61/0.73 = 3.5(2) = 7$$



7. A compound in a soft drink that has 0.31g hydrogen, 3.2g phosphorus, and 6.5g oxygen.

$$\text{H} - 0.31/1 = 0.31/.1 = 3.1$$

$$\text{P} - 3.2/31 = 0.10/.1 = 1$$

$$\text{O} - 6.5/16 = 0.40/.1 = 4$$



Empirical And Molecular Formula Quiz With Answers

Mrinal Talukdar



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This book delves into Empirical And Molecular Formula Quiz With Answers. Empirical And Molecular Formula Quiz With Answers is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Empirical And Molecular Formula Quiz With Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Empirical And Molecular Formula Quiz With Answers
 - Chapter 2: Essential Elements of Empirical And Molecular Formula Quiz With Answers
 - Chapter 3: Empirical And Molecular Formula Quiz With Answers in Everyday Life
 - Chapter 4: Empirical And Molecular Formula Quiz With Answers in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of Empirical And Molecular Formula Quiz With Answers. This chapter will explore what Empirical And Molecular Formula Quiz With Answers is, why Empirical And Molecular Formula Quiz With Answers is vital, and how to effectively learn about Empirical And Molecular Formula Quiz With Answers.
 3. In chapter 2, this book will delve into the foundational concepts of Empirical And Molecular Formula Quiz With Answers. The second chapter will elucidate the essential principles that need to be understood to grasp Empirical And Molecular Formula Quiz With Answers in its entirety.
 4. In chapter 3, the author will examine the practical applications of Empirical And Molecular Formula Quiz With Answers in daily life. This chapter will showcase real-world examples of how Empirical And Molecular Formula Quiz With Answers can be effectively utilized in everyday scenarios.
 5. In chapter 4, this book will scrutinize the relevance of Empirical And Molecular Formula Quiz With Answers in specific contexts. This chapter will explore how Empirical And Molecular Formula Quiz With Answers is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, this book will draw a conclusion about Empirical And Molecular Formula Quiz With Answers. This chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Empirical And Molecular Formula Quiz With Answers.

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