

### Assignment #3 – Compound Names and Formulas Single and Multi-valent ions

A. Write the correct formula for the following compounds.

1. gallium oxide  $Ga_2O_3$
2. mercury (I) chloride  $HgCl$
3. tin (IV) bromide  $SnBr_4$
4. indium oxide  $In_2O_3$
5. cadmium sulphide  $CdS$
6. magnesium oxide  $MgO$
7. potassium nitride  $K_3N$
8. iron (III) sulphide  $Fe_2S_3$
9. gold (III) bromide  $AuBr_3$
10. zinc oxide  $ZnO$

11. silicon oxide  $SiO_2$
12. aluminum bromide  $AlBr_3$
13. tin (II) sulphide  $SnS$
14. germanium bromide  $GeBr_4$
15. nickel (II) oxide  $NiO$
16. strontium astatide  $SrAt_2$
17. aluminum oxide  $Al_2O_3$
18. copper (II) oxide  $CuO$
19. lawrencium fluoride  $LrF_3$
20. actinium oxide  $Ac_2O_3$

B. Name the following compounds correctly. Only include the Roman Numeral in the name if the first element has 2 or more combining capacities.

1.  $CaBr_2$  Calcium Bromide
2.  $K_2O$  Potassium Oxide
3.  $Fe_2O_3$  Iron (III) Oxide
4.  $ZnF_2$  Zinc Fluoride
5.  $CuCl$  Copper (I) Chloride
6.  $Ni_2O_3$  Nickel (III) Oxide
7.  $AuCl$  Gold (I) Chloride
8.  $HgBr$  Mercury (I) Bromide

9.  $AsCl_5$  Arsonic (V) Chloride
10.  $Eu_2O_3$  Europium (III) Oxide
11.  $NdBr_3$  Neodymium Bromide
12.  $Cu_3N_2$  Copper (II) Nitride
13.  $CoS$  Cobalt (II) Sulphide
14.  $ZrO_2$  Zirconium Oxide
15.  $Li_2O$  Lithium Oxide
16.  $NaH$  Sodium Hydride

# Naming Covalent Compounds Answer Sheet

**SBPD Editorial**



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