Predicting Reaction Products and Naming

Predict the products for the following reactions, balance the equation, classify the type of reaction and then name the reactants and products. Be sure to include the states for double displacement reactions

- Ca(OH)₁ + ____ HCl →
- 2) ____ Pb(NO₃)₂ + ____ K₂CrO₄ →
- 3) ____ NaC₁H₂O₂ + ____ H₂SO₃ →
- 4) ____ Cu(OH)₁ + ____ H₂PO₃ →
- 5) ____ NH₄Be + ____ Ca(OH)₂ →
- 6) ____ Na + ___ FeBr₁ →
- 7) ____NaOH + ____H₂SO₄ →
- 8) ____ C₂H₄O₂ + ____ limiting O₂ →
- 9) ____ NH₁ + ____ H₂O →
- 10) ____ PbSO₄ + ____ AgNO₃ →
- 11) _____ AllBr₂ →
- 12) ____ HBr + ___ Fe →
- 13) ____ KClO₁ + ____ ZnCl₂ →
- 14) ____MnO₂ + ____ Sn(OH)₄ →
- 15) ____ excess O₂ + ___ C₃H₁₃O₂ →
- 16) ____ H₂O₂ →
- 17) ____ KNO₂ →

Naming Chemical Reactions Practice

Christopher R. Hren, John T. Moore, Peter J. Mikulecky

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