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Assignment #2 – Compound Names and Formulas Multi-valent ions only

A. Write the correct formula for the	e following compounds, all of which have been named using the
	The ionic charge is given after the first element (metallic) in Roman
Numerals.	

1. copper (II) oxide CUO	11. manganese (III) oxide My 203		
2. mercury (I) oxide	12. vanadium (II) bromide VBrz		
3. gold (III) chloride AuCl3	13. niobium(V) oxide Nb≥05		
4. thallium (III) bromide TIBY 3	14. titanium (III) oxide T\20>		
5. bismuth (V) oxide 81205	15. titanium (III) nitride TIN		
6. terbium (III) oxide	16. iron (II) oxide <u>FeO</u>		
7. uranium (VI) oxide \\(\lambda \)	17. cobalt (II) phosphide C03P2		
8. protactinium (IV) bromide PhBY4	18. tin (II) oxide		
9. cerium (III) oxide	19. thulium (II) bromide TWBY 2		
10. arsenic (V) sulphide A5255	20. copper (II) bromide CUBY2		
B. Name the compound using the Roman Numeral Method and balance the ionic charges.			
1. Snc4 Tin (IV) Chloride	9. Pdf. Palladium (IV) Fluoride		
2. BIBIO BISMINHY (V) Bromide	10. Os2O3 OSMIUM (III) Oxide		
3. POO2 ROLONIUM (IV) OXICE	11. MOBEZ MOLYLOGENIUM (II) Bramide		
4. Pbl2 150d(II) Iodide	12. VCIs Vanadium (V) Chloride		
5. HgO <u>Hercury</u> (II) Oxide	13. MnzO3 Manganese (III) Oxide		
6. Hgci <u>Hercury</u> (I) Chroride	14. COO COLONH (II) Oxide		
7. Augos Gold (III) Oxide	15. NP2O3 NEDTUNIUM (II) OXKIC		
8. Fecto Iron (II) Chloride	16. V2Os Vanadium (V) Oxide		

Naming Balancing And Formula Writing Answer Key

William L. Masterton, Emil J. Slowinski, Conrad L. Stanitski

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