# Naming Ionic Compounds - Answer Key

Give the name of the following ionic compounds:

### Name

1)	Na <sub>2</sub> CO <sub>3</sub>	sodium carbonate
2)	NaOH	sodium hydroxide
3)	MgBr <sub>2</sub>	magnesium bromide
4)	KCI	potassium chloride
5)	FeCl <sub>2</sub>	iron (II) chloride
6)	FeCl <sub>3</sub>	iron (III) chloride
7)	Zn(OH) <sub>2</sub>	zinc hydroxide
8)	Be <sub>2</sub> SO <sub>4</sub>	beryllium sulfate
9)	CrF <sub>2</sub>	chromium (II) fluoride
10)	$AJ_2S_3$	aluminum sulfide
11)	PbO	lead (II) oxide
12)	Li <sub>3</sub> PO <sub>4</sub>	lithium phosphate
13)	Titla	titanium (IV) iodide
14)	Co <sub>3</sub> N <sub>2</sub>	cobalt (II) nitride
15)	Mg <sub>3</sub> P <sub>2</sub>	magnesium phosphide
16)	Ga(NO <sub>2</sub> ) <sub>3</sub>	gallium nitrite
17)	Ag <sub>2</sub> SO <sub>3</sub>	silver sulfite
18)	NH <sub>4</sub> OH	ammonium hydroxide
19)	AI(CN) <sub>3</sub>	aluminum cyanide
20)	Be(CH <sub>3</sub> CO	O) <sub>2</sub> beryllium acetate

# Naming Ionic Compounds Practice Worksheetanswer Key

**Laurie E. Westphal** 

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web dec 10 2019 i have a free live online facilitator guide template or lesson plan that you can download and use for your webinar and virtual classroom planning and delivery in this blog post i am going to look through the history of my lesson plans over the years to highlight the good practice and different options available to you