

AICE Marine Science AS



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Chapter 1 Notes – Quiz 1.2 Solubility in water

Answer Key /25

1. Seawater is a mixture of solutes and solvents. Seawater should be considered a homogeneous solution.

2a) Define the term solute.
A substance that dissolves in a solvent

2b) Describe how sodium chloride dissolves in water.
Through dissolution of ions
Partially positive hydrogen of water molecules attracts the negative chloride ions
Partially negative oxygen and δ^- attracts the positive sodium ion and the δ^- from the Cl^- ions

3a) Solubility of sodium chloride can be shown.
 Temperature impacts the solubility of sodium chloride.
As water heats up, the amount of ions found in sodium chloride increases in relation to the solution, rather than just the water.

3b) The average salinity of seawater is 35 ppt. Describe how this body of water.

- Through natural evaporation
- Will 1000 g of seawater until all water evaporates and the salt is left

3c) Explain how evaporation would impact salinity.

- Salinity will rise
- As seawater evaporates faster, water is lost faster than sodium and other particles helping an increase in salinity

2. The diagram below illustrates changes in dissolved oxygen levels from surface to the bottom of the water.

3a) State the name of the area of extremely low oxygen.
Hypoxia

3b) Describe the trends in dissolved oxygen levels, as seen in the diagram.
As water depth increases, the dissolved oxygen levels decrease. The oxygen levels are highest at the surface and lowest at the bottom.

3c) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
At the surface, there is a high concentration of oxygen. As depth increases, the oxygen levels decrease. This is due to the fact that oxygen is consumed by organisms and the rate of diffusion is slower at greater depths.

3d) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
At the surface, there is a high concentration of oxygen. As depth increases, the oxygen levels decrease. This is due to the fact that oxygen is consumed by organisms and the rate of diffusion is slower at greater depths.

3e) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
At the surface, there is a high concentration of oxygen. As depth increases, the oxygen levels decrease. This is due to the fact that oxygen is consumed by organisms and the rate of diffusion is slower at greater depths.

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At the surface, there is a high concentration of oxygen. As depth increases, the oxygen levels decrease. This is due to the fact that oxygen is consumed by organisms and the rate of diffusion is slower at greater depths.

3i) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
At the surface, there is a high concentration of oxygen. As depth increases, the oxygen levels decrease. This is due to the fact that oxygen is consumed by organisms and the rate of diffusion is slower at greater depths.

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At the surface, there is a high concentration of oxygen. As depth increases, the oxygen levels decrease. This is due to the fact that oxygen is consumed by organisms and the rate of diffusion is slower at greater depths.

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At the surface, there is a high concentration of oxygen. As depth increases, the oxygen levels decrease. This is due to the fact that oxygen is consumed by organisms and the rate of diffusion is slower at greater depths.

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At the surface, there is a high concentration of oxygen. As depth increases, the oxygen levels decrease. This is due to the fact that oxygen is consumed by organisms and the rate of diffusion is slower at greater depths.

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3x) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
At the surface, there is a high concentration of oxygen. As depth increases, the oxygen levels decrease. This is due to the fact that oxygen is consumed by organisms and the rate of diffusion is slower at greater depths.

3y) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
At the surface, there is a high concentration of oxygen. As depth increases, the oxygen levels decrease. This is due to the fact that oxygen is consumed by organisms and the rate of diffusion is slower at greater depths.

3z) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
At the surface, there is a high concentration of oxygen. As depth increases, the oxygen levels decrease. This is due to the fact that oxygen is consumed by organisms and the rate of diffusion is slower at greater depths.

Date	1 st ppt measurement	2 nd ppt measurement	3 rd ppt measurement	Average ppt
September 7	8.1			
September 8	8.0	8.2		
September 9	8.1	7.9	8.3	8.1
		7.8	8.0	8.0
			8.1	8.0

3a) State what ppt is.
A measure of the concentration of hydrogen ions in a solution.

3b) Describe the method to measure ppt level in a solution.
Using a pH meter or indicator. The pH meter would be the most accurate way to measure pH. The indicator would be less accurate as it only gives a range of pH values.

3c) Describe a method to measure the pH level in a solution.
Using a pH meter or indicator. The pH meter would be the most accurate way to measure pH. The indicator would be less accurate as it only gives a range of pH values.

3d) Calculate the average ppt level for September 8.
Average ppt = $\frac{8.2 + 7.9 + 8.3}{3} = 8.1$

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RJ Shavelson



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An Introduction to the World's Oceans Keith A. Sverdrup, Alyn C. Duxbury, Alison Duxbury, 2003 In An Introduction to the World's Oceans Seventh Edition Keith Sverdrup Alyn Duxbury and Alison Duxbury have blended the most contemporary information and research with basic principles to bring you and your students an unmatched comprehensive introduction to oceanography You will find a significantly revised Seventh Edition that addresses all the latest findings in oceanography What's special about these authors An Introduction to the World's Oceans Seventh Edition contains balanced and comprehensive coverage that comes from each author having strength in different areas of oceanography Oceanography is an eclectic science that examines physical chemical and biological properties of the world's oceans Alison Duxbury has a background in marine biology Alyn Duxbury has a background in physical oceanography and Keith Sverdrup has a background in marine geology geophysics and how oceanography relates to other areas of science The result A well balanced comprehensive introduction to oceanography McGraw Hill has exclusive videos from Scripps Institution of Oceanography These video clips will be brief one to two minute clips and available on either videotape or on the Digital Content Manager CD ROM There will be a total of about 2 hours and 12 minutes worth of these short clips Clips will be available for each chapter of the text and no other company can offer these videos

Resources in Education, 1995

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Citizen Science for Coastal and Marine Conservation John A. Cigliano, Heidi L. Ballard, 2017-10-31 In recent years citizen science has emerged as a powerful new concept to enable the general public students and volunteers to become involved in scientific research A prime example is in biodiversity conservation where data collection and monitoring can be greatly enhanced through citizen participation This is the first book to provide much needed guidance and case studies from marine and coastal conservation The novelty and rapid expansion of the field has created a demand for the discussion of key issues and the development of best practices The book demonstrates the utility and feasibility as well as limitations of using marine and coastal citizen science for conservation and by providing critical considerations i.e. which questions and systems are best suited for citizen

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provides a systematic review of nature based solutions and their potential to address current environmental challenges In the 21st century society is faced by rapid urbanisation and population growth degradation and loss of natural capital and associated ecosystem services an increase in natural disaster risks and climate change With growing recognition of the need to work with ecosystems to resolve these issues there is now a move towards nature based solutions which involve utilising nature s ecosystem to solve societal challenges while providing multiple co benefits This book systematically reviews nature based solutions from a public policy angle assessing policy developments which encourage the implementation of nature based solutions to address societal challenges while simultaneously providing human well being and biodiversity benefits This includes enhancing sustainable urbanisation restoring degraded ecosystems mitigating and adapting to climate change and reducing risks from natural disasters While nature based solutions can be applied strategically and equitably to help societies address a variety of climatic and non climatic challenges there is still a lack of understanding on how best to implement them The book concludes by providing a best practice guide for those aiming to turn societal challenges into opportunities This book will be of great interest to policymakers practitioners and researchers involved in nature based solutions sustainable urban planning environmental management and sustainable development generally

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