

A crystalline solid is formed when 2 or more atoms exchange valence electrons to form an extended stability.

Ionic bond is formed when valence electrons are transferred from one atom to another.

cations = atoms lose electrons to become [+] charged cations.

anions gain electrons to become [-] charged.

In ionic compounds that ions are arranged in a crystal lattice. Strong forces hold the ions together.

Properties of ionic compounds:

- = high melting and boiling points
- = hard & not easily crushed
- = conduct electricity when melted or dissolved because the ions are free to move

covalent bond = electrons are shared. Forming molecules

Covalent compounds have weak forces holding the molecules together.

Properties of covalent compounds:

- low melting and boiling points
- Many covalent compounds are volatile liquids or gases.
- soft, easier to crush
- are not conductors of electricity
- electronegativity = property that tells how strong an

Gbp Episode 6answers

N Colangelo



Gbp Episode 6 answers:

Reviewing **Gbp Episode 6answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Gbp Episode 6answers**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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