Equation Writing and Predicting Products

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If you examine a car that has been in a junkyard for a while, you will notice that it has rusted. Rusting in a slow chemical reaction of the iron to the ear's body with oxygen gas. If a piece of sodium is put into water, a much more rapid chemical reaction occurs. Sodium reacts with water to produce sodium bydroxide and bydrogen gas. During this reaction, enough heat is liberated to ignite the hydrogen gas, causing it to explode.

Chemists observe what is happening in a chemical reaction and try to describe it in language that is simple and clear. A chemical equation uses formulas and symbols to describe the substances involved in a reaction, the physical state of the substance, the use of a catalyst, and relative proportions. The general form of an equation is:

Reactants --- Preducts

In this investigation you will perform a series of reactions and make careful observations of the changes that occur. Using simple tests and your knowledge of chamistry, you will describe the identity of the products. With this information, you will write chamical equations to describe the reactions.

Pro-Eab Discussion:

bulanced chemical equation.

Read through the entire lab investigation and the relevant pages of your textbook. Then answer the questions that hillow.

	a. oxygen b. bydregen c. carbon dioxide
	d. water vapor e. seramonia (NH ₂)
=	What is the proper way to smell a substance in the lab? Why should care be taken when smelling it gas such as attraconia?
3.	What is the role of a establish in a reaction? How can you tell when a substance serves as a catalyst? (This answer will be important latait, spend some time on it now.)
a.	One way to identify limestone (CaCO ₃) is to drop a small amount of hydrochloric acid (HCI) on it. A positive test results in a first of carbon dioxide being

produced. Write out in words the information represented by the following

CaCCuto + 2 HCl taxt - + CaCL(aq) + HcO(1) + CO(ca)

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