

- (A) sodium carbonate
(C) alum

- (B) Sodium sulphate
(D) Sodium chloride

CHEMISTRY

2012

Time: 2 Hours 40 Minutes

Marks: 68

SECTION 'B' (SHORT-ANSWER QUESTIONS)

NOTE: Attempt any 5 questions from Inorganic & 5 questions from Organic Chemistry. (40)

INORGANIC CHEMISTRY

- 2.(i) State the Modern periodic law. Explain how it removed the defects of Mendeleev's System of classification.
- (ii) (a) Write the Valence shell configuration of the following groups. * IA and IB * VA and VB
(b) What is Aqua Regia? How does gold dissolve in it? Give equations for the reaction.
- (iii) Give the reason of the following:
* Ionic Hydrides are called true Hydrides.
* Lithium and Beryllium markedly differ from other members of their respective groups.
* Plaster of Paris is used in making plaster coats and moulds.
* Why is the electronic configuration of Chromium (Cr) $4s^1, 3d^5$ instead of $4s^2, 3d^4$ while that of Copper (Cu) is $4s^1, 3d^{10}$ instead of $4s^2, 3d^9$?
- (iv) Give equations for the following:
* Zinc is treated with 90% conc H_2SO_4
* Action of Heat on $KMnO_4$
* Aluminum reacts with aqueous Sodium Hydroxide
* Lunar caustic is heated $450^\circ C$
- (v) (a) Write a note on any one: Detergents OR Tin plating.
(b) Write I.U.P.A.C. names of the following:
* $[Fe(NO_2)_6]^{3-}$ * $[Cr(en)_2 Cl_2]Cl$
- (vi) Describe the preparation of Chlorine gas by Castner-Kellner cell or Nelson cell.
- (vii) Complete the following equations and give I.U.P.A.C. names of the complexes formed:
* $CuSO_4 + NH_3 \xrightarrow{\text{excess}}$ * $AgNO_3 + NH_3 \xrightarrow{\text{excess}}$
- (viii) (a) What is Blue vitriol? Give two equations for its preparation.
(b) What are outer and inner transition elements and how many series of each of them are present in the periodic table? Also write the names of the inner transition series.

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1. Understanding the eBook January 2f32chemistry Past Paper
 - The Rise of Digital Reading January 2f32chemistry Past Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying January 2f32chemistry Past Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an January 2f32chemistry Past Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from January 2f32chemistry Past Paper

- Personalized Recommendations
- January 2f32chemistry Past Paper User Reviews and Ratings
- January 2f32chemistry Past Paper and Bestseller Lists
- 5. Accessing January 2f32chemistry Past Paper Free and Paid eBooks
 - January 2f32chemistry Past Paper Public Domain eBooks
 - January 2f32chemistry Past Paper eBook Subscription Services
 - January 2f32chemistry Past Paper Budget-Friendly Options
- 6. Navigating January 2f32chemistry Past Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - January 2f32chemistry Past Paper Compatibility with Devices
 - January 2f32chemistry Past Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of January 2f32chemistry Past Paper
 - Highlighting and Note-Taking January 2f32chemistry Past Paper
 - Interactive Elements January 2f32chemistry Past Paper
- 8. Staying Engaged with January 2f32chemistry Past Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers January 2f32chemistry Past Paper
- 9. Balancing eBooks and Physical Books January 2f32chemistry Past Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection January 2f32chemistry Past Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine January 2f32chemistry Past Paper
 - Setting Reading Goals January 2f32chemistry Past Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of January 2f32chemistry Past Paper

- Fact-Checking eBook Content of January 2f32chemistry Past Paper
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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