

## **Naming Ionic Compounds – Answer Key**

*Give the name of the following ionic compounds:*

|     |                               | <b>Name</b>                   |
|-----|-------------------------------|-------------------------------|
| 1)  | $\text{Na}_2\text{CO}_3$      | <b>sodium carbonate</b>       |
| 2)  | $\text{NaOH}$                 | <b>sodium hydroxide</b>       |
| 3)  | $\text{MgBr}_2$               | <b>magnesium bromide</b>      |
| 4)  | $\text{KCl}$                  | <b>potassium chloride</b>     |
| 5)  | $\text{FeCl}_2$               | <b>iron (II) chloride</b>     |
| 6)  | $\text{FeCl}_3$               | <b>iron (III) chloride</b>    |
| 7)  | $\text{Zn(OH)}_2$             | <b>zinc hydroxide</b>         |
| 8)  | $\text{Be}_2\text{SO}_4$      | <b>beryllium sulfate</b>      |
| 9)  | $\text{CrF}_2$                | <b>chromium (II) fluoride</b> |
| 10) | $\text{Al}_2\text{S}_3$       | <b>aluminum sulfide</b>       |
| 11) | $\text{PbO}$                  | <b>lead (II) oxide</b>        |
| 12) | $\text{Li}_3\text{PO}_4$      | <b>lithium phosphate</b>      |
| 13) | $\text{TiI}_4$                | <b>titanium (IV) iodide</b>   |
| 14) | $\text{Co}_3\text{N}_2$       | <b>cobalt (II) nitride</b>    |
| 15) | $\text{Mg}_3\text{P}_2$       | <b>magnesium phosphide</b>    |
| 16) | $\text{Ga(NO}_2)_3$           | <b>gallium nitrite</b>        |
| 17) | $\text{Ag}_2\text{SO}_3$      | <b>silver sulfite</b>         |
| 18) | $\text{NH}_4\text{OH}$        | <b>ammonium hydroxide</b>     |
| 19) | $\text{Al(CN)}_3$             | <b>aluminum cyanide</b>       |
| 20) | $\text{Be(CH}_3\text{COO)}_2$ | <b>beryllium acetate</b>      |

# Formulas Of Ionic Compounds Answer Key

**Rachel S Tattersall**



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