

## 14.3

## HYDROGEN IONS AND ACIDITY

## Section Review

## Objectives

1. Calculate concentrations based on the pH of acids and bases and vice versa.
2. Calculate pH and concentrations based on  $K_a$  and  $K_b$  and vice versa.
3. Calculate the pH of weak acids and bases.

## Vocabulary

- |                   |                             |                        |
|-------------------|-----------------------------|------------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. acid solution       |
| 4. autoionization | 5. acid rain                | 6. acid-base indicator |
|                   | 7. acid solution            | 8. pH                  |

## Key Equation

- |                                     |                                     |
|-------------------------------------|-------------------------------------|
| 1. $\text{pH} = -\log [\text{H}^+]$ | 2. $\text{pH} = -\log [\text{H}^+]$ |
| 3. $\text{pH} = -\log [\text{H}^+]$ | 4. $\text{pH} = -\log [\text{H}^+]$ |

## Part A Completion

Write the equation and/or the numerical value(s) of the unknown quantity that are indicated in the space provided. Be sure to include units where appropriate.

1. What substance is \_\_\_\_\_, contains hydrogen ions?
2. \_\_\_\_\_ is the concentration of those acids present in rainwater that equal to \_\_\_\_\_.
3. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
4. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
5. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
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8. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
9. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
10. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
11. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
12. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
13. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
14. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
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16. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
17. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
18. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
19. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.
20. \_\_\_\_\_ is the concentration of those acids present in \_\_\_\_\_.

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# Hydrogen Ions And Acidity Section Review Answers

**Kevin T. Patton, PhD**



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**Publications** United States. National Bureau of Standards,1969      **Publications of the National Bureau of Standards** United States. National Bureau of Standards,1968      *Publications of the National Bureau of Standards, 1968-1969* United States. National Bureau of Standards,Betty L. Oberholtzer,1970      **Catalog of National Bureau of Standards Publications, 1966-1976: pt. 1 Citations and abstracts. v. 2. Key word index (A through L)** United States. National Bureau of Standards. Technical Information and Publications Division,1978      NBS Special Publication ,1968

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