BACKS TOWNSON, S NIV 30 5 x 1 M = 0.5479 M MEAN, IN I BUTHE WANTED

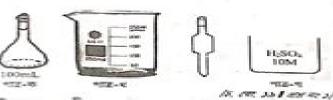
Stee we, w - 20 m HICH HIS SEMINE WILL M = 36.5 g/medHarvis minor, $\nabla = 11$. Herefore represents $N \ll \gamma$

্রা কুলাকে বালিও X, লাকাভিতে 10 mg এক কেছে কম পরিমাণ দ্বাৰ $_{\rm B}$ $_{
m V}$ -symblectic $_{
m 0.5}$ g $_{
m 200}$ $\left(\frac{M}{V}\right)$ 1 K.1 Weyer Stein mins and strains : কুলা X. পৃথাতিটি হলে সেমি হাইকো আন্দানাইটিকাল পৃথাতি ও ত্রতার হলো মাতো আনগাইটিকাল ককতি। উত্তর কতিত্র মুক্তিকাল ককটি পরিবেশ কক্ষেত্র চিক্ত ত্ব প্ৰতিয়োজন কৰিছিল কৰিছেল আৰুৰ। নিয়ে তা বিয়েছণ কৰা

ক্রমেইরের পাবতি রামায়নিক বুগগত বিরেছদের একটি পাবতি। ক্রমেইরের পাবতি রামায়নিক বুগগত বিরেছদের একটি পাবতি। ুত বেছামাটার বসায়নে মহা পরিমাশ রামায়নিক উপাদনে স্নায়করণ ও ক্রিলেকে ডিমেবল বাবস্থত হয়। এ পশ্বতিতে সুব তার পরিমাণ (0.05 ত্রত (i.2g তর্তুন বা 2-4m), তরণ প্রবা) রাস্ক্রেনিক পদার্থ নিছে ্তিত করা হয়। কলে পতিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আদে। সেমি মাইজো পাবাতিহত 111,58 बार शरिकार्ड তার অনুসামাইত ব্যবহার করা হয় কলে পরিবেশ দূষণ ঘটে না। CH.CSNH; + H₂O ---- CH₂CONH₂ + H₂S

_{লিকা} খোনো কোনো কেতে কম ভূকির রাসায়নিক দুবা বাবস্তুত বয়। কলে পরিবেশ দুঘণের সম্ভাবনা কমে আলে। তাই সেমি-মাইজ্রে পশ্বতি কটোৱেল কলেবার ।

ELLES



- ত, তাইভার প্রথক কীপু
- সেমি-মাইক্রো পশ্বতি পরিবেশ বান্ধর কেন?
- ্ ত্রণটির জন্য প্রযোজ্য ম্যাজার্ড প্রতীকটি অচকন করে। এবং यहारता स्ट्रहा ।
- ্ উন্নীপকের আলোকে দ্রবদ থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে IM ঘনমাত্রার প্রবণ তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪

৩ নং প্রয়োর উত্তর

🌌 বিশেষণীয় নিস্থির বীমের উপর রাইভার স্থাপন করলে বীমের প্রতি ^{মান্তির} কন্য যে ভর পাওয়া যায়, তাকে রাইভার ধুবক বলে।

🔯 সেমিনইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। ্ট বিচেম্পির রসায়নে ভয় পরিমাপ রাসায়নিক উপাদান সনাজকরণ ও র্থনিন্দত বিশ্লেষ্ট্র ব্যবস্থৃত হয়। এ পন্ধতিতে দুব অয় পরিমাণ ^{(0,05} সেকে 0.2g কঠিন বা 2-4ml, তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীয়া করা হয়। ফলে পরিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আসে। ^{ছাবন্ত}, কোনো কোনো কেন্তে কম ঝুঁকির রাসায়নিক মধ্য ব্যবহৃত হয়। ^{মাস পরিবেশ} দূরণের সম্ভাবনা কমে আসে। তাই সেমি-মাইরেন পশ্বতি পরিবেশ সংস্থাব ।

ট্র ইমিশরের প্রবর্গটি হলো সালফিউরিক এপিড (H₂SO₄) এর 10 M ইল। H₅SO₄ একটি জাত্রক ও কয়কারী পদার্থ। H₂SO₄ এর 10 M ইল হাৰা পাচ H₂SO₄ স্তবণ। গাঢ় H₂SO₄ পদাৰ্থকে ক্ষ্যাকাৰ্ত্তী আজাৰ্ত তিত্বৈ অনুসূত্র করা হয়। গাড় H2SO4 এর জন্য করকারী হাজার হটার নিয়ে অন্তব্দ করে ব্যাখ্যা করা হলো:

er-libra	
	weeds (Corresive)
कार,न्हर्स	ভয়বারী রাজ্যানিক উপাদান করতে যে সকল রাম্যানিক চ্বাকে বুবার যার সংস্পর্য বুক ও অন্যান্য
Beaute	পান্ধার্থক মানাছক কৃতি সাহিত হয়। এতে তার পুরু ভার এবং সন্মানা পানর পরে হার। এতে তার পুরু পারে করিল ক্রিকার্ড্র (HCI, HNO), HISO), পার করে (Nicht, Roll), H ₂ O ₂ , আমেরিয়া ক্রিকার বিশ্বিয়া কর্ম
PERMIT	The state of the s
শতর্জত	শায় কৰিছেকে বোজকে আৰতে হৰে। শায় কৰেকে বাৰতে ন্যালয়ত্ব বিকাৰক বোজকে আৰতে হৰে।

ট্রালাকের মুনশাটি হতের 1604 H₅SO, প্রদান এবং এর বেকে 1M ঘনমান্ত্ৰত প্ৰথম প্ৰায়ত কৰাত মাল ১৯ ৫ ৩৪ কাড্ৰয় বাংলাৰ কৰাত

১৭ কাচয়ন্ত্ৰটি হলো 100 ml. আহাতনমিতিক ভ্ৰাণক ও এই টি হলো পিলেট। যেমেতু আয়তনমিতিক য়াংশেরর আয়তন 100 mt., তাই আমানের 100 mL IM খনমাত্রার H₋SO, প্রবদ তৈরি করতে হবে : দ্রবংগর অধুকরণ সূত্র যতে আমরা লাগি,

$$V_1S_1 = V_2S_2$$

 $V_1 = \frac{V_2S_2}{S_1}$
 $= \left(\frac{100 \times 1}{10}\right) \text{ mL}$
 $= 10 \text{ mL}$

আৰ্থমিক আত্তৰ, ৮, – ৫ $484121, S_1 = 10M$ CON SCHOOL V2 = 100 mL weight, $S_2 = 1M$.

অক্তর্যর, 10mL H₃SO, স্তব্যের সচেদ পানি যোগ করতে মনে (100-10) mL = 90 mL

আবার, শিপেটের স্মায়ায়ে 10 mL প্রদান খনমাত্রার প্রবণ নিচে তার সাবে আয়তনমিতিক ফ্লান্ডে 90 mL পানি যোগ করলে আমালের কাজিনত খনমাত্রার দ্রবণ পাওয়া যাবে ৷

EH COS







58-A

Per Line Day II with the Art

- ক, আন্টিঅক্সিডেট কাকে বলেং
- थ. माथा धूरक ८, धर यान कथाना अभीय रह गा–साथा PERCENT I
- A ভিত্তর মন্ত্রটিকে উত্তর করার কৌশল বর্ণনা করাে।
- R ত C চিছের সাথে সম্পর্কিত রাসায়নিক প্রবাসির ব্যবহারকালে দ্বাস্থ্য কৃতি স্তাদে করণীয় উপায় সম্পর্কে विद्रायम् करता ।

ह पर कार्यात किन्द्र

আটিঅভিডেউ এক ধরনের রাসারনিক বা কৃত্রিম খাদা শবেকক য়া মূলত থাৰা উপাদানসমূহকে জাৱণ প্ৰতিষ্কায় তাৱন থেকে ৱকা করে থাবার সংরক্ষণ করে।

সাম্যাতক K, এর মান কথনো অধীম বা পূন্য হতে পারে মা। উত্তিটি স্থাখ্যা করার জন্য নিচের উতমুখী সমীকরণটি বিবেচনা করি— $A + B \rightleftharpoons C + D$

Hsc Chemistry 1 First Chapter Notes

Joacim Rocklöv

Hsc Chemistry 1 First Chapter Notes:

Excel HSC Chemistry Jim Stamell, 2011 ISBN 9781741252996 AUTHOR Jim Stamell RRP 39 95 PAGES 428 pp SPECIFICATION Softcover perfect bound 280 mm x 210 mm STATUS New edition PUBLICATION DATE April 2008 The EXCEL HSC Chemistry guide is directly linked to the syllabus with every sin gle dot point of the HSC Chemistry syllabus appearing in the margin of the book You can write in the guide so your study is focused and your notes are structured This guide comes in a brand new format that makes even better use of your study time up to date covera ge of the core topics plus 3 Option topics Industrial Chemistry Shipwr ecks Corrosion and Conservation and Forensic Chemistry this guide is organised just like the HSC syllabus so the students learn to s ection the theoretical part is under routine headings and the students section the practical part is under headings like First hand Second h and Investigations and Problem Solving %this way you will be ab le to see at a glance what the theoretical and practical work is all main headings in each chapter 1 1 2 1 etc are directly from the syllabus word for word %this way you can easily match the Excel guide to the syllabus an alphabetical list of all the key definitions and concepts you should know from each chapter %an efficient way of learning all the definitions in one go chapter syllabus checklist with every single dot point listed in checklist form for each chapter %a fantastic way of testing that you know all the work hundreds of key concept questions with answers %questions that test you recall of knowledge in each chapter HSC type quest ions for every section in each chapter with clock icons to tell you how much time you will have to answer the questions in the HSC %this way yo u can test yourself on HSC type questions under HSC type time pressure an examiner maximiser feature ticks to show the mark distribut ion and answers to all HSC type questions %all you need to answ er HSC type questions two sample HSC papers with an examiner m aximiser feature plus answers %not one but two up to date sample papers the Excel syllabus summary notes a detachable section at the end of the guide where every single dot point of each chapter is summarised for you% a comprehensive and compact summary of the whole course in 32 pages The American Journal of Science ,1911 The academy ,1890 "The" Athenaeum .1897

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