

# GRAPHING TRENDS OF THE PERIODIC TABLE

## CHEMISTRY

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PERIOD: \_\_\_\_\_

### Graphing Periodic Table Trends

**Directions:** Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	152	Sodium - 11	190	Potassium - 19	243	Rubidium - 37	265
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 38	219
Boron - 5	87	Aluminum - 13	175	Gallium - 31	136	Indium - 49	156
Carbon - 6	67	Silicon - 14	111	Germanium - 32	125	Tin - 50	140
Nitrogen - 7	56	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	115

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 37	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 38	550

### Atomic Radius



# Graphing Periodic Table Trends 1 Answers

**Ann Fullick, Patrick Fullick**



## Graphing Periodic Table Trends 1 Answers:

Basic Concepts of Chemistry Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. WileyPLUS sold separately from text.

**Chemistry in the Community.** American Chemical Society, 2002 This volume has relevance to a wide number of courses giving a hands-on introduction to chemistry in relation to community issues rather than around specific chemical concepts.

**Chemistry in the Community (Enhanced Core Four)** American Chemical Society, 2006-02-15

**Jacaranda Chemistry 1 VCE Units 1 and 2, learnON and Print** Neale Taylor, Angela Stubbs, Robert Stokes, 2022-11-30 Developed by expert Victorian teachers for VCE students. The NEW Jacaranda Chemistry VCE series continues to deliver curriculum-aligned material that caters to students of all abilities. Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design 2023/2027.

Chemistry in the Community (ChemCom) American Chemical Society, 2011-06-17 Touted as the most successful NSF-funded project published, Chemistry in the Community ChemCom by the American Chemical Society (ACS) offers a meaningful and memorable chemistry program for all levels of high school students. ChemCom covers traditional chemistry topics within the context of societal issues and real-world scenarios. Centered on decision-making activities where students are responsible for generating data, investigating, analyzing that data, and then applying their chemistry knowledge to solve the presented problem. The text is intensively laboratory-based with all 39 of the investigations integrated within the text, not separate from the reading. With the ChemCom program, students learn more organic and biochemistry, more environmental and industrial chemistry, and more on the particulate nature of matter than other textbooks, all within the relevance of solving problems that arise in everyday life. Meticulously updated to meet the needs of today's teachers and students, the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards. Incorporating advances in learning and cognitive sciences, ChemCom's wide-ranging coverage builds upon the concepts and principles found in the National Science Education Standards. Correlations are available showing how closely aligned ChemCom is to these and other state standards.

*Chemistry for AQA.* Ann Fullick, Patrick Fullick, 2001 This resource has separate books for biology, chemistry, and physics. Each book is accompanied by a teacher's resource pack on customizable

CD ROM or as a printed pack The series is designed to work in conjunction with the Coordinated Science for AQA series so that coordinated and separate science can be taught alongside each other      **Prentice Hall Science Explorer** Michael J. Padilla,2002      **Holt Chemistry** Ralph Thomas Myers,2004      **Sif Chemistry NI Twb 2e** Rex M. Heyworth,2007

**Chemistry: The Central Science** Theodore L. Brown,H. Eugene LeMay Jr.,Bruce E. Bursten,Catherine Murphy,Patrick Woodward,Steven Langford,Dalius Sagatys,Adrian George,2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation      The Science Teachers Bulletin ,1989      **Science Units for Grades 9-12** Randy L. Bell,Joe Garofalo,2005 Sample topics include cell division virtual dissection earthquake modeling the Doppler Effect and more      *AQA A Level Chemistry (Year 1 and Year 2)* Alyn G. McFarland,Nora Henry,Teresa Quigg,2019-07-22 Develop and learn to apply your knowledge progressing from basic concepts to more complicated Chemistry with worked examples practical activities and mathematical support in this updated all in one textbook for Years 1 and 2 Written for the AQA A level Chemistry specification this revised textbook will Provide support for all 12 required practicals with activities that introduce practical work and other experimental investigations in Chemistry Offer detailed examples to help you get to grips with difficult concepts such as physical chemistry calculations Helps to improve mathematical skills with support throughout examples of method and a dedicated Maths for chemistry chapter Allow you to easily measure progression with differentiated end of topic questions and Test yourself questions Develop understanding with free online access to Test yourself answers Practice question answers and extended glossaries      Chemistry Homework for OCR A for Double and Separate Awards Gareth Pritchard,2001 Howework activites for OCR A Chemistry specifications      Advanced Chemistry for You Lawrie Ryan,2000 Advanced Chemistry for You has been carefully designed to be interesting and motivating to the AS A2 student with features that make it highly supportive of individual learning Written by an experienced author team with the same straightforward approach as the successful New for You GCSE series      **Instructor's Resource Manual** Silberberg,1999-07      *Journal of Research of the National Bureau of Standards* United States. National Bureau of Standards,1988      **Prentice Hall Chemistry** ,2000      *Johnstone Triangle* Norman Reid,2021-01-18 Chemistry is often seen as a difficult subject to understand This book focusses on the triangle model that Alex H Johnstone developed in the early 1980s Originally conceived in the context of making chemistry more accessible to a wider range of learners the model

has been applied in almost every area of education in chemistry at all stages of learning In looking at why chemistry is difficult there are two central questions Firstly does the problem relate to the nature of chemistry and secondly does it relate to the way humans gain understanding Both were found to be important and the answers to the two question were found to be connected The triangle model arose from sustained research into human learning The central finding from research is the critical role of working memory and the model rationalises so much evidence from chemistry education research as well as the repeated experiences of teachers of chemistry at all levels In order to understand chemistry it is essential to develop sound mental models of molecular reality It generates major implications for the way a chemistry curriculum should be constructed and the processes of teaching and learning in chemistry when the goal is focussed on understanding the key ideas Some of these implications are developed and pointers offered to more successful ways forward The power of the Johnstone Triangle lies in the way it offers clear directions for all involved in chemistry education It is hoped that this book will prove helpful to all involved in sharing the exciting story of the way humans have come to understand the molecular world one of the great examples of great human endeavour

**MCAT General Chemistry Review 2023-2024** Kaplan Test Prep, 2022-08-02 Kaplan's MCAT General Chemistry Review 2023 2024 offers an expert study plan detailed subject review and hundreds of online and in book practice questions all authored by the experts behind the MCAT prep course that has helped more people get into medical school than all other major courses combined Prepping for the MCAT is a true challenge Kaplan can be your partner along the way offering guidance on where to focus your efforts and how to organize your review This book has been updated to match the AAMC's guidelines precisely no more worrying about whether your MCAT review is comprehensive The Most Practice More than 350 questions in the book and access to even more online more practice than any other MCAT general chemistry book on the market The Best Practice Comprehensive general chemistry subject review is written by top rated award winning Kaplan instructors Full color 3 D illustrations from Scientific American charts graphs and diagrams help turn even the most complex science into easy to visualize concepts All material is vetted by editors with advanced science degrees and by a medical doctor Online resources including a full length practice test help you practice in the same computer based format you'll see on Test Day Expert Guidance High yield badges throughout the book identify the topics most frequently tested by the AAMC We know the test The Kaplan MCAT team has spent years studying every MCAT related document available Kaplan's expert psychometricians ensure our practice questions and study materials are true to the test

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