

Stoichiometry - Gram Conversions

1. Use the following balanced equation to answer the questions below:



- How many grams of O_2 are produced when 34.5g of HgO decompose?
 - How many grams of HgO need to decompose to produce 5.27g of Hg ?
 - If 52.1g of HgO decomposes, then how many grams of Hg will be made?
2. When catalyzed by sulfuric acid, potassium chlorate and sucrose react to produce heat, CO_2 , H_2O , and KCl . Below is a balanced equation for this reaction.



- How many grams of CO_2 will form if 52.9g of sucrose react completely with KClO_3 ?
 - How many grams of KClO_3 are needed to completely react with 781g of $\text{C}_{12}\text{H}_{22}\text{O}_{11}$?
 - How many moles of water will be produced if 99.1g of KClO_3 reacts completely with sucrose?
3. Nitrogen dioxide reacts with water to produce nitric acid and nitric oxide. Below is the balanced equation for this reaction.



- How many grams of NO will be made if 5.57g of NO_2 reacts completely with water?

Gram To Gram Stoichiometry Worksheet Answers

Jennifer Louise Pakkala



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