CASCALLER AND PROPERTY.

(There is no Prelaboratory Assignment for this exercise. Make entries in your lab notebook for anything that is an observation; for your report, turn in this page and the following page.)

# Making molecular models

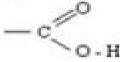
Make models of the following functional groups (additional red O balls will be needed):

make one C-C group

make four -H

make two C=C groups

make four



groups

Make models of the following dicarboxylic acids and observe:

- A. Rotation around chemical bonds:
  - Make succinic acid, HOOC-CH2-CH2-COOH.

Does the molecule rotate around the central C-C bond?

2.Replace the central CH2-CH2 group with a HC=CH group (keep one COOH group on each end).

Does the molecule rotate around the central C=C bond?

- 3. What is the molecular formula (C<sub>i</sub>H<sub>j</sub>O<sub>k</sub>)of your new molecule?
- B. Comparison of maleic and fumaric acids.

Maleic acid and fumaric acid have the same molecular formula (C<sub>4</sub>H<sub>4</sub>O<sub>4</sub>) and the same condensed structural formula (HOOC-CH=CH-COOH). Your model is either maleic or fumaric acid.

- Compare your model with others in the class. What differences do you see?
- 2. Make a second model of the other form.

# **Geometric Isomers Lab Answers**

**Padhraic Smyth** 

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