

Assignment #2 – Compound Names and Formulas
Multi-valent ions only

A. Write the correct formula for the following compounds, all of which have been named using the modern Roman Numeral Method. The ionic charge is given after the first element (metallic) in Roman Numerals.

1. copper (II) oxide CuO
2. mercury (I) oxide Hg₂O
3. gold (III) chloride AuCl₃
4. thallium (III) bromide TlBr₃
5. bismuth (V) oxide Bi₂O₅
6. terbium (III) oxide Tb₂O₃
7. uranium (VI) oxide UO₃
8. protactinium (IV) bromide PaBr₄
9. cerium (III) oxide Ce₂O₃
10. arsenic (V) sulphide As₂S₅

11. manganese (III) oxide Mn₂O₃
12. vanadium (II) bromide VBr₂
13. niobium(V) oxide Nb₂O₅
14. titanium (III) oxide Ti₂O₃
15. titanium (III) nitride TiN
16. iron (II) oxide FeO
17. cobalt (II) phosphide Co₃P₂
18. tin (II) oxide SnO
19. thulium (II) bromide TmBr₂
20. copper (II) bromide CuBr₂

B. Name the compound using the Roman Numeral Method and balance the ionic charges.

1. SnCl₄ Tin (IV) Chloride
2. BiBr₅ Bismuth (V) Bromide
3. PoO₂ Polonium (IV) Oxide
4. PbI₂ Lead (II) Iodide
5. HgO Mercury (II) Oxide
6. HgCl Mercury (I) Chloride
7. Au₂O₃ Gold (III) Oxide
8. FeCl₂ Iron (II) Chloride

9. PdF₄ Palladium (IV) Fluoride
10. Os₂O₃ Osmium (III) Oxide
11. MoBr₂ Molybdenum (II) Bromide
12. VCl₅ Vanadium (V) Chloride
13. Mn₂O₃ Manganese (III) Oxide
14. CoO Cobalt (II) Oxide
15. Np₂O₃ Neptunium (III) Oxide
16. V₂O₅ Vanadium (V) Oxide

Formulae Of Ionic Compounds Answer Key

Jianjun Gao



Formulae Of Ionic Compounds Answer Key:

Decoding **Formulae Of Ionic Compounds Answer Key**: Revealing the Captivating Potential of Verbal Expression

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Table of Contents Formulae Of Ionic Compounds Answer Key

1. Understanding the eBook Formulae Of Ionic Compounds Answer Key
 - The Rise of Digital Reading Formulae Of Ionic Compounds Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Formulae Of Ionic Compounds Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Formulae Of Ionic Compounds Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Formulae Of Ionic Compounds Answer Key
 - Personalized Recommendations

- Formulae Of Ionic Compounds Answer Key User Reviews and Ratings
- Formulae Of Ionic Compounds Answer Key and Bestseller Lists
- 5. Accessing Formulae Of Ionic Compounds Answer Key Free and Paid eBooks
 - Formulae Of Ionic Compounds Answer Key Public Domain eBooks
 - Formulae Of Ionic Compounds Answer Key eBook Subscription Services
 - Formulae Of Ionic Compounds Answer Key Budget-Friendly Options
- 6. Navigating Formulae Of Ionic Compounds Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Formulae Of Ionic Compounds Answer Key Compatibility with Devices
 - Formulae Of Ionic Compounds Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Formulae Of Ionic Compounds Answer Key
 - Highlighting and Note-Taking Formulae Of Ionic Compounds Answer Key
 - Interactive Elements Formulae Of Ionic Compounds Answer Key
- 8. Staying Engaged with Formulae Of Ionic Compounds Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Formulae Of Ionic Compounds Answer Key
- 9. Balancing eBooks and Physical Books Formulae Of Ionic Compounds Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Formulae Of Ionic Compounds Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Formulae Of Ionic Compounds Answer Key
 - Setting Reading Goals Formulae Of Ionic Compounds Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Formulae Of Ionic Compounds Answer Key
 - Fact-Checking eBook Content of Formulae Of Ionic Compounds Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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