

Explain how you identified the limiting and excess reactant based only on your qualitative observations (not quantitative measurements or calculations). Type your answer into the box. Use complete sentences.

According to observation the initial color of bromocresol purple is yellow in color after adding the NaOH, it turns violet. After Changing the color yellow from purple indicates acid is completely consumed so excess of base addition changes the color into violet.

Table C. Calculations to Determine Theoretical Yields

<i>Description</i>	<i>Calculated Value</i>
Volume of NaOH solution	22.20 mL
Initial moles of NaOH	0.002451 moles
Initial moles of H_3PO_4	0.002602 moles
Mass of H_2O produced if NaOH runs out first	0.04414 g
Mass of H_2O produced if H_3PO_4 runs out first	1.265 g
Theoretical yield of H_2O (in grams)	1.265 g
Theoretical yield of Na_3PO_4 (in grams)	0.4267 g

Table D. Identification of Limiting and Excess Reactants from Calculations

<i>Description</i>	<i>Reactant Name or Formula</i>
Limiting Reactant	Phosphoric Acid
Excess Reactant	Sodium Hydroxide

Explain how you identified the limiting and excess reactant based on your calculations. Be specific. Type your answer into the box. Use complete sentences.

According to our calculation the reacting equivalent of H_3PO_4 is less than NaOH. Therefore H_3PO_4 is limiting reactant so product formation depends on H_3PO_4 because phosphoric acid is completely consumed.

Formal Lab Report On Limiting Reactant

C Cleary



Formal Lab Report On Limiting Reactant:

The book delves into Formal Lab Report On Limiting Reactant. Formal Lab Report On Limiting Reactant is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Formal Lab Report On Limiting Reactant, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Formal Lab Report On Limiting Reactant
- Chapter 2: Essential Elements of Formal Lab Report On Limiting Reactant
- Chapter 3: Formal Lab Report On Limiting Reactant in Everyday Life
- Chapter 4: Formal Lab Report On Limiting Reactant in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Formal Lab Report On Limiting Reactant. This chapter will explore what Formal Lab Report On Limiting Reactant is, why Formal Lab Report On Limiting Reactant is vital, and how to effectively learn about Formal Lab Report On Limiting Reactant.
3. In chapter 2, this book will delve into the foundational concepts of Formal Lab Report On Limiting Reactant. This chapter will elucidate the essential principles that need to be understood to grasp Formal Lab Report On Limiting Reactant in its entirety.
4. In chapter 3, the author will examine the practical applications of Formal Lab Report On Limiting Reactant in daily life. This chapter will showcase real-world examples of how Formal Lab Report On Limiting Reactant can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Formal Lab Report On Limiting Reactant in specific contexts. This chapter will explore how Formal Lab Report On Limiting Reactant is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Formal Lab Report On Limiting Reactant. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Formal Lab Report On Limiting Reactant.

https://staging.conocer.cide.edu/data/detail/Documents/Lear_Siegler_Overhaul_Manual.pdf

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