Polyatomic Ions

Can a group of atoms have a charge?

Why?

Do you know you eat a lot of "-ates"? Next time you look at a food label, read the ingredients and you will likely find a number of ingredients that end with "-ate," such as sodium phosphate or calcium carbonate. Did you ever wonder what the chemical formulas of these ingredients look like? In this activity we will explore polyatomic ions, which are groups of atoms that carry a charge. These ions are found in our food ingredients, natural waterways, and many other chemical compounds you encounter every day.

Model 1 - Types of Ions

| Monatomic Ions | Nitride N | Sulfide S | Chloride Cl)1- | |
|--------------------|-----------------|--------------|-------------------|--|
| Polyatomic Ions | Nitrate | Sulfate | Ammonium H N H | |
| Jons | Nitrite Nitrite | Sulfite | Hydroxide | |

1. Use Model 1 to complete the table below.

| Name of-lon | Nitride | Nitrate | Sulfate | Sulfite | Ammonium |
|--------------------------------|------------|------------------------|----------------------|----------------------|--------------------------|
| Charge on Ion | -3 | -1 | -2 | -2 | +1 |
| Type and Number of Atoms | I nitrogen | 1 nitrogen 3 oxygen | 1 sulfur 4 oxygen | 1 sulfur 3 oxygen | 1 nitrogen 4 hydrogen |
| Chemical Formula | N3- | NO,1- | SO ₄ - | SO,2- | NH,'· |

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K. Kuchitsu

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Structure of Free Polyatomic Molecules K. Kuchitsu, Kozo Kuchitsu, 1998-09-10 This volume Structure of Free Polyatomic Molecules Basic Data contains frequently used data from the corresponding larger Landolt B rnstein handbooks in a low price book for the individual scientists working in the laboratory Directories link to the more complete volumes in the library The book contains important information about a large number of semiconductors Structure Data of Free Polyatomic Molecules K. Kuchitsu, 1995-11-27 Since the publication of Volumes II 7 in 1976 and its supplements II 15 in 1987 and II 21 in 1992 the information on the structure of free molecules in the ground state and in excited electronic states has increased considerably Therefore this volume II 23 contains data from 148 inorganic and 498 organic polyatomic free molecules including free radicals and molecular ions published between 1990 and 1993 inclusively and a small number of structures published 1994 All experimental methods for the determination of structural data of free molecules have been considered all data obtained by these methods have been critically evaluated and compiled The structural data for more than 3400 polyatomic free molecules can be completely surveyed and easily retrieved by means of this volume

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