

Specific Heat Problems – ANSWERS

The last page of this document has Miscellaneous Information that you will need for these problems.

1. How many calories are needed to warm 55.8 g of water at 21.3°C to 73.4°C?

$$\text{Heat} = m \times S. H. \times (T_{\text{final}} - T_{\text{initial}})$$

$$\text{Heat} = 55.8 \text{ g} \times 1.00 \text{ cal/g}^\circ\text{C} \times (73.4^\circ\text{C} - 21.3^\circ\text{C})$$

$$\text{Heat} = 55.8 \text{ g} \times 1.00 \text{ cal/g}^\circ\text{C} \times 52.1^\circ\text{C}$$

$$\text{Heat} = 2907.18 \text{ cal} \quad \text{convert to scientific notation}$$

$$= 2.9\textbf{0}718 \times 10^3 \text{ cal}$$

$$= 2.91 \times 10^3 \text{ cal (3 sig. figs.)}$$

2. How many calories are needed to warm 55.8 g of alcohol from 21.3°C to 73.4°C?

$$\text{Heat} = m \times S. H. \times (T_{\text{final}} - T_{\text{initial}})$$

$$\text{Heat} = 55.8 \text{ g} \times 0.58 \text{ cal/g}^\circ\text{C} \times (73.4^\circ\text{C} - 21.3^\circ\text{C})$$

$$\text{Heat} = 55.8 \text{ g} \times 0.58 \text{ cal/g}^\circ\text{C} \times 52.1^\circ\text{C}$$

$$\text{Heat} = 1686.16 \text{ cal} \quad \text{convert to scientific notation}$$

$$\text{Heat} = 1.6\textbf{8}616 \times 10^3 \text{ cal}$$

$$\text{Heat} = 1.69 \times 10^3 \text{ cal}$$

Heat Practice Problems Answers

Léo Pomar

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Heat Practice Problems Answers:

Whispering the Strategies of Language: An Psychological Quest through **Heat Practice Problems Answers**

In a digitally-driven world wherever screens reign great and quick transmission drowns out the subtleties of language, the profound secrets and emotional nuances hidden within words often get unheard. However, situated within the pages of **Heat Practice Problems Answers** a fascinating literary value blinking with organic thoughts, lies an extraordinary journey waiting to be undertaken. Penned by a talented wordsmith, this enchanting opus attracts readers on an introspective journey, gently unraveling the veiled truths and profound affect resonating within the cloth of each word. Within the emotional depths of the emotional review, we can embark upon a genuine exploration of the book is core styles, dissect its captivating writing fashion, and yield to the powerful resonance it evokes heavy within the recesses of readers hearts.

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