## KEY!

## Chemistry 12 - Review of Significant Digits

TOTAL (28)

The rules for zeros in significant digits are as follows:

- a) All zeros between non-zero digits are significant.
- b) Zeros at the <u>beginning of a number</u> (eg. 0.0095) are NOT SIGNIFICANY! If the number 0.0095 was written in scientific notation, it would be: 9.5 x 10<sup>-3</sup>. The exponent is not counted as significant so this number has 2 significant digits.
- Zeros on the <u>right side</u> of a number (at the end) are significant if the DECIMAL POINT is shown.
  - eg) 50.00 has 4 significant digits 43.0 has 3 significant digits 20, has 2 significant digits 100, has 3 significant digits
- Zeros to the left of an UNDERSTOOD decimal point are NOT significant.
  - eg) 300 has I significant digit 10 000 has I significant digit 12 320 has 4 significant digits 420 has 2 significant digits
- Find the number of significant digits in each of the following measurements:

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**D** Keegan

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