

Notes on Study-Session

Changes in Kinetics:

Rate of Reaction:

Rate of Consumption: $\Delta[\text{Reactant}]$

Rate of Change: $\Delta[\text{Product}]$

Rate of Accumulation: $\text{Accumulation Rate} = \text{Rate of Production} - \text{Rate of Consumption}$

Rate-Independent Effects are:

- temperature phase diagram
- partition coefficient calculations
- partition volume calculations
- distribution coefficients with partition equations
- temperature scaling rules
- ideal & non-ideal solution diagrams

Rate-Dependent:

(1) Concentration-Dependent Rate:

Reaction Kinetics: - the study of how changes in time affect the concentration of chemical species

Order: - description of how concentration affects the rate of reaction

Rate: - a physical property of a substance

Order: - the rate of reaction

Reaction: - the physical change involved in chemical reaction

Concentration: - the amount of substance present

Concentration: - the amount of substance present

Order: - the amount of substance present in a reaction

Concentration: - the amount of substance present in a reaction

(2) Effect of Temperature on the Rate of Reaction:

Order:

(3) Effect of the amount of substance on the rate of reaction:

Order:

(4) Effect of the amount of substance on the rate of reaction:

Order:

(5) Effect of the amount of substance on the rate of reaction:

Order:

(6) Effect of the amount of substance on the rate of reaction:

Order:

(7) Effect of the amount of substance on the rate of reaction:

Order:

Myers Unit 10 Study Guide Answers

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