

QUESTION 4

4.1 Calculate the missing amounts (indicated by a, b and c) in the Fixed/Tangible Asset Note for the year ended 28 February 2014.

NO.	WORKINGS	AMOUNT
a	6 740 000 – 2 689 000 no part marks	4 051 000 ✓✓
b	140 000 – 33 600 no part marks	106 400 ✓✓
c	$5\,600 + 11\,000 + 20\% \times (3\,110\,000 - 1\,342\,000)$ <p style="text-align: center;">353 600 (3 marks)</p> <p>OR:</p> $5\,600 + 20\% \times \left[\frac{1\,768\,000}{10} \times \frac{10}{12} + \frac{2\,098\,000}{10} \times \frac{4}{12} \right]$ <p style="text-align: center;">370 200 (6 marks)</p>	one part correct 370 200 <input checked="" type="checkbox"/>

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4.2 METEOR SUPERMARKETS LTD
CASH FLOW STATEMENT FOR YEAR ENDED 28 FEBRUARY 2014

CASH FLOW STATEMENT FOR YEAR ENDED 31 FEBRUARY 2014		
Cash flow from operating activities	(1 039 900)	<input checked="" type="checkbox"/> Operational
Cash generated from operations	969 600	
Interest paid	(88 500)	
✓ Taxation 441 000 ✓ + 122 000 ✓ + 128 000 ✓ Details & correctly placed 563 000 (2 marks) OR -441 000 - 122 000 - 128 000	(691 000) 691 000: 3 marks	<input checked="" type="checkbox"/> Operation; one part correct & brackets
✓ Dividends 480 000 ✓ + 750 000 ✓ ✓ Details & correctly placed 480 000 + 1 520 000 - 870 000 OR - 480 000 - 1620 000 + 870 000 1 mark 1 mark 1 mark	(1 230 000) 1 230 000: 3marks	<input checked="" type="checkbox"/> Operation one part correct & brackets
Cash flow from investing activities	(4 274 600)	<input checked="" type="checkbox"/> Operational
✓ Purchase of fixed assets 4 051 000 <input checked="" type="checkbox"/> + 330 000 ✓ Details & correctly placed see 4.1 a	(4 381 000) 4 381 000: 2 marks	<input checked="" type="checkbox"/> Operation one part correct & brackets
✓ Proceeds from disposal of fixed assets Details & correctly placed	106 400	<input checked="" type="checkbox"/> See 4.1b & no brackets
Cash flow from financing activities	2 325 000	<input checked="" type="checkbox"/> Operational
✓ Proceeds from the sale of shares Details & correctly placed	2 400 000 ✓ 2 400 000: 1 mark	<input checked="" type="checkbox"/> correct & no brackets
✓ Share buy-back 50 000 x R9.50 Details & correctly placed	(475 000) ✓ 475 000: 1 mark	<input checked="" type="checkbox"/> correct & brackets
✓ Change in loan Details & correctly placed	400 000 ✓ 400 000: 1 mark	<input checked="" type="checkbox"/> correct & no brackets
Net change in cash and cash equivalents	(2 989 500)	<input checked="" type="checkbox"/> Operation
Cash and cash equivalents at beginning of year	2 488 000	
Cash and cash equivalents at end of year	(501 500)	✓ <input checked="" type="checkbox"/> *

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* Operation from Sep 98 B13 500 98 /B515 900/

Valid alternative details acceptable # Operation one part correct in the sub-section (figures/details)

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G Orfield



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by its rate law for example the rate of the gas phase decomposition of dinitrogen pentoxide $2\text{N}_2\text{O}_5 \rightarrow 4\text{NO}_2 + \text{O}_2$ has been found to be directly proportional to the concentration of N_2O_5 text rate k n

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web the rate determining step limits the overall rate and therefore determines the rate law for the overall reaction

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web the initial rate law is then measured for each of the reactions consider the reaction between nitrogen monoxide gas and hydrogen gas to form nitrogen gas and water vapor the following data were collected for this reaction at 1280 text o text c

see table below notice that the starting concentrations of ce no and ce h

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web determination of a rate law and activation energy in this experiment glass beakers pipettes and a hotplate were utilized to set up the procedures a solution of deionized water a buffer potassium iodide ki sodium thiosulfate $\text{na}_2\text{s}_2\text{o}_3$ starch and hydrogen peroxide h_2o_2 were used in order to calculate the rate of reaction

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web rate laws sometimes called differential rate laws or rate equations are mathematical expressions that describe the relationship between the rate of a chemical reaction and the concentration of its reactants as an example consider the reaction described by the chemical equation $\text{a} + \text{b} \rightarrow \text{products}$

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web determining the rate law page 1 page 2 page 3 kinetics experiments the goal of a kinetics experiment is to measure the concentration of a species at a particular time during a reaction so that a rate law can be determined

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