

10.1 Periodic Trends



10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1: Introduction

In your notebook, read aloud the modern periodic table.

Copy the letter-headers in the first row on the left below from the periodic table to complete the table on the right.

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

- _____ 10. It varies on the periodic table.
- _____ 11. It has no line periodic table.
- _____ 12. Group 1 elements.
- _____ 13. Elements that are strong oxidizing agents.
- _____ 14. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 15. There are two representative elements of elements.
- _____ 16. Most elements have higher of the elements in the periodic table are metals.
- _____ 17. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 18. Group 1A elements are the most reactive metals.
- _____ 19. Group 1A elements are highly reactive nonmetals known as halogens.
- _____ 20. Group 1A elements are very reactive elements known as representative metals.
- _____ 21. Alkaloids have properties of both metals and nonmetals.

Hydrocarbons Study Guide Answers For Content Mastery

Daniela Niemeyer



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