

Hydrocarbons

Methane	General formula for an alkane	General formula for an alkene	Ethene	Propene	Butene
Ethane	How to test between an alkene and alkane	Alkane or alkene? C_2H_6 C_3H_8 C_4H_{10} C_5H_{12}	Show how ethene is polymerised		
Propane	Definition of a hydrocarbon		What is meant by saturated/unsaturated		
Butane	How ethanol is produced... ...from ethane		Draw the monomer that makes this polymer $ \begin{array}{cccccc} \text{Br} & \text{H} & \text{Br} & \text{H} & \text{Br} & \text{H} \\ & & & & & \\ \text{H}-\text{C}-\text{C}-\text{C}-\text{C}-\text{C}-\text{C}-\text{H} \\ & & & & & \\ \text{H} & \text{H} & \text{H} & \text{H} & \text{H} & \text{H} \end{array} $		
A way to remember the order of alkanes	...by fermentation		Describe the process of cracking Process:		Complete the equations for cracking $C_{10}H_{22} \rightarrow C_2H_6 + \text{_____} + C_6H_{14}$ $C_8H_{18} \rightarrow C_3H_8 + \text{_____}$
			Types of products produced:		The types of products produced:
			Conditions:		
			Why it is done:		

Hydrocarbon Compounds Core Teaching Resources

Michael Brown



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