

# Periodic Table Basics

**Step 1:** Complete the square for each element by filling in the atomic number, name, & atomic mass.

**Step 2:** Determine the number of protons, neutrons, and electrons in an atom of each element.




**Step 3:** Identify if the element is a solid, liquid, or gas at room temperature.

**Step 4:** Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

**Step 5:** List at least three physical or chemical properties for each element.

**Step 6:** List at least three uses for each element.

**Step 7:** Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

|   |   |                   |
|---|---|-------------------|
|    | P = <br>N = <br>E =  | O S<br>O L<br>O G |
|   | M.P. = <br>B.P. =   |                   |
| Properties   |   |                   |
| Uses   |   |                   |
| Bohr Diagram    Lewis Structure   |   |                   |
|    |   |                   |

**Step 8:** Use the following colors to shade in the square for each element. You should ONLY color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na  
Orange = B & Al

Pink = O & S  
Red = C & Si

Blue = Be & Mg  
Tan = N & P

Purple = F & Cl  
Yellow = He, Ne, & Ar

**Step 9:** Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

|    |    |    |    |    |    |    |    |
|----|----|----|----|----|----|----|----|
| 1  |    |    |    |    |    |    | 2  |
| 3  | 4  | 5  | 6  | 7  | 8  | 9  | 10 |
| 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 |

Done? Answer the questions on the worksheet using the information on your Periodic Table!

# Heat Calculations Physical Science If8767 Answers

**L Towne**



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