

Laboratory Exercise: Double Replacement Reactions...

TABLE II: SCALE UP OF PRECIPITATION REACTIONS

Row #	Chemical Formula of Reagent	Chemical Formula of Reagent	List the four ions present in reagents		insoluble	soluble
					Formula of Solid Product appearance in test tube	Formula of Aqueous Product appearance in test tube
2	NaOH	FeCl ₃	Na ⁺	OH ⁻	Fe(OH) ₃	
			Fe ⁺³	Cl ⁻	very small orange	very cloudy orange
3	NaNO ₃	FeCl ₃	Na ⁺	NO ₃ ⁻		
			Fe ⁺³	Cl ⁻	solid white powder	yellow cloudy
5	Na ₂ CO ₃	FeCl ₃	Na ⁺	CO ₃ ⁻²		
			Fe ⁺³	Cl ⁻	solid orange powdery	very cloudy orange
6	Na ₃ PO ₄	FeCl ₃	Na ⁺	PO ₄ ⁻³		
			Fe ⁺³	Cl ⁻	yellowish small	yellow cloudy
12	AgNO ₃	NaOH				
					not almost brown	cloudy brown
13	NaOH	CuSO ₄	N			
					gel like chunk	blue uncent
17	AgNO ₃	Na ₂ CO ₃				
					white flake small	white cloudy
18	AgNO ₃	Na ₃ PO ₄				
					yellow egg	clear yellow
19	CuSO ₄	Na ₂ CO ₃				
					small blue/white powder	cloudy blue
20	CuSO ₄	Na ₃ PO ₄				
					very small blue particles	cloudy light-blue

Lab 9 2 Double Replacement Reaction

IM Harris



Lab 9 2 Double Replacement Reaction:

Reviewing **Lab 9 2 Double Replacement Reaction**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Lab 9 2 Double Replacement Reaction**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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