

Equilibrium Lab

Instructions: Complete all sections of this report and answer all questions in complete sentences for full credit.

Title:

Objective(s):

Hypothesis: *Complete the following statements for your hypotheses. Choose one of the options in brackets by highlighting, bolding, or placing your answer in a different color.*

If the amount of reactant increases, the rate of the forward reaction will [**increase** or decrease] and the rate of the reverse reaction will [increase or **decrease**].

If the amount of reactant decreases, the rate of the forward reaction will [increase or **decrease**] and the rate of the reverse reaction will [**increase** or decrease].

If the amount of product increases, the rate of the forward reaction will [increase or **decrease**] and the rate of the reverse reaction will [**increase** or decrease].

If the amount of product decreases, the rate of the forward reaction will [**increase** or decrease] and the rate of the reverse reaction will [increase or **decrease**].

Materials: For this lab, you are not expected to list the materials. But still notice the materials (lab equipment and chemicals) used in the lab to help you analyze the data, write the conclusion, and answer the post-lab questions.

Variables: *List variables below as you work through the simulation.*

| Type | Variables |
|-------------|---------------------------|
| Controlled | Equilibrium |
| Independent | How much of the reactants |
| Dependent | Reaction rate |

Procedure: *Follow the steps described in the virtual lab. Record your observations and interpretations below.*

Equilibrium reaction in the lab:



Data: *For each trial, do the following:*

Lab 3disturbing Equilibrium Answers

Clemens Wendtner

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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on September 16 18 1974 This book is organized into four sessions encompassing 19 chapters The opening session deals with the emulsification process and emulsion polymerization as well as the adsorption behavior of polyelectrolyte stabilized emulsions The following session examines the rheological properties stability and fluid mechanics of emulsions This session also looks into the role of protein conformation and crude oil water interfacial properties in emulsion stability The third session highlights the preparation formation properties and application of bitumen emulsions The concluding session describes the process of spontaneous emulsification the steric emulsion stabilization the interfacial measurements of oil in water emulsions and the influence of the disperse phase on emulsion stability This book will be of value to chemists chemical and process engineers and researchers

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 - Chapter 5: Conclusion
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