

Name \_\_\_\_\_ date \_\_\_\_\_

**Ideal Gas Law Worksheet Review**

$$PV=nRT$$

mass  $\longleftrightarrow$  mole

$$R = \begin{array}{l} 0.0821 \text{ L}\cdot\text{atm/mol}\cdot\text{K} \\ 62.4 \text{ L}\cdot\text{mm Hg/mol}\cdot\text{K} \\ 8.314 \text{ L}\cdot\text{kPa/mol}\cdot\text{K} \end{array}$$

Standard pressure and temperature (STP): 1 atm and 273 K

---

1. How many moles of a gas at 100°C does it take to fill a 1.00 L flask to a pressure of 1.50 atm?
2. A camping stove holds 3000 g of liquid  $\text{C}_3\text{H}_8$ . How large of a container would be needed to hold the gas at a temperature of at 25°C and a pressure of 3.0 atm? (hint: need to convert to moles first)
3. What volume would be occupied by 3.5 mol of oxygen gas at a pressure of 768 mm Hg and a temperature of 25°C?
4. A ten-liter gas cylinder contains 55 moles nitrogen. What pressure, in kPa, is exerted by the nitrogen at 25°C?
5. A drum used to transport crude oil has a volume of 162 L. How many grams of water, as steam, are required to fill the drum at 1.20 atm and 100°C?
6. How many moles of air are there in a 125 mL flask if the pressure is 755 mm Hg and the temperature is 20°C?

# Ideal Gas Law Worksheet With Answers

**CO Houle**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'CO Houle' text.

## **Ideal Gas Law Worksheet With Answers:**

**A Guide to Microsoft Excel 2013 for Scientists and Engineers** Bernard Liengme, 2015-03-17 Completely updated guide for students scientists and engineers who want to use Microsoft Excel 2013 to its full potential Electronic spreadsheet analysis has become part of the everyday work of researchers in all areas of engineering and science Microsoft Excel as the industry standard spreadsheet has a range of scientific functions that can be utilized for the modeling analysis and presentation of quantitative data This text provides a straightforward guide to using these functions of Microsoft Excel guiding the reader from basic principles through to more complicated areas such as formulae charts curve fitting equation solving integration macros statistical functions and presenting quantitative data Content written specifically for the requirements of science and engineering students and professionals working with Microsoft Excel brought fully up to date with the new Microsoft Office release of Excel 2013 Features of Excel 2013 are illustrated through a wide variety of examples based in technical contexts demonstrating the use of the program for analysis and presentation of experimental results New to this edition The Backstage is introduced a new Office 2013 feature all the external operations like Save Print etc are now in one place The chapter on charting is totally revised and updated Excel 2013 differs greatly from earlier versions Includes many new end of chapter problems Most chapters have been edited to improve readability **Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Physics Workbook For Dummies** Steven Holzner, 2007-10-05 Do you have a handle on basic physics terms and concepts but your problem solving skills could use some static friction Physics Workbook for Dummies helps you build upon what you already know to learn how to solve the most common physics problems with confidence and ease Physics Workbook for Dummies gets the ball rolling with a brief overview of the nuts and bolts i e converting measures counting significant figures applying math skills to physics problems etc before getting into the nitty gritty If you re already a pro on the fundamentals you can skip this section and jump right into the practice problems There you ll get the lowdown on how to take your problem solving skills to a whole new plane without ever feeling like you ve been left spiraling down a black hole With easy to follow instructions and practical tips Physics Workbook for Dummies shows you how to you unleash your inner Einstein to solve hundreds of problems in all facets of physics such as Acceleration distance and time Vectors Force Circular motion Momentum and kinetic energy Rotational kinematics and rotational dynamics Potential and kinetic energy Thermodynamics Electricity and magnetism Complete answer explanations are included for all problems so you can see where you went wrong or right Plus you ll get the inside scoop on the ten most common mistakes people make when solving physics problems and how to avoid them When push comes to shove this friendly guide is just what you need to set your physics problem solving skills in motion **Unique Scientific Puzzles** Dr. S. Pancharatnam, 2020-04-06 Born and brought up in a sugar factory village Pancharatnam turned into a good scholar in leading school and college in Pune then IIT Bombay This propelled him into some of the world s best

universities UC Berkeley and Stanford He managed to get away from the ivory towers of USA and devote to more challenging and rewarding Indian chemical industry with more useful RD and project engineering Another success story was his own business of specialty filters for the mech engineering industry with over hundred reputed customers So here he is with a fully enjoyable career of fifty years with over fifty projects and many publications Having spent all his life in technical investigation and improvements he has brought to you vast variety of 500 interesting puzzles from various fields Most are actually encountered in daily life Many are truly unique and some quite advanced Further over 500 jokes are added for relaxing in between So go ahead struggle laugh and learn a lot This small book is highly recommended for students of final years of school all college students in science engineering and tech professionals Even teachers will find it interesting for setting tests Of course riddles and easy puzzles can be enjoyed by everyone Free quarterly updates are assured on your email id for 1 year

**A Guide to Teaching in the Active Learning Classroom** Paul Baepler, J. D. Walker, D. Christopher Brooks, Kem Saichaie, Christina I. Petersen, 2023-07-03 While Active Learning Classrooms or ALCs offer rich new environments for learning they present many new challenges to faculty because among other things they eliminate the room's central focal point and disrupt the conventional seating plan to which faculty and students have become accustomed The importance of learning how to use these classrooms well and to capitalize on their special features is paramount The potential they represent can be realized only when they facilitate improved learning outcomes and engage students in the learning process in a manner different from traditional classrooms and lecture halls This book provides an introduction to ALCs briefly covering their history and then synthesizing the research on these spaces to provide faculty with empirically based practical guidance on how to use these unfamiliar spaces effectively Among the questions this book addresses are How can instructors mitigate the apparent lack of a central focal point in the space What types of learning activities work well in the ALCs and take advantage of the affordances of the room How can teachers address familiar classroom management challenges in these unfamiliar spaces If assessment and rapid feedback are critical in active learning how do they work in a room filled with circular tables and no central focus point How do instructors balance group learning with the needs of the larger class How can students be held accountable when many will necessarily have their backs facing the instructor How can instructors evaluate the effectiveness of their teaching in these spaces This book is intended for faculty preparing to teach in or already working in this new classroom environment for administrators planning to create ALCs or experimenting with provisionally designed rooms and for faculty developers helping teachers transition to using these new spaces

**Fundamentals of Analytical Chemistry** Douglas A. Skoog, 2004 This text is known for its readability combined with a systematic rigorous approach Extensive coverage of the principles and practices of quantitative chemistry ensures suitability for chemistry majors

*Backpacker*, 2001-03 Backpacker brings the outdoors straight to the reader's doorstep inspiring and enabling them to go more places and enjoy nature more often The authority on active adventure Backpacker is the world's

first GPS enabled magazine and the only magazine whose editors personally test the hiking trails camping gear and survival tips they publish Backpacker's Editors Choice Awards an industry honor recognizing design feature and product innovation has become the gold standard against which all other outdoor industry awards are measured **Popular Mechanics**

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**Most Asked Questions on Ideal Gas Law - What You Need to Know** Philip Hicks,2014-10-07 It's a brand new Ideal gas law world There has never been a Ideal gas law Guide like this It contains 51 answers much more than you can imagine comprehensive answers and extensive details and references with insights that have never before been offered in print Get the information you need fast This all embracing guide offers a thorough view of key knowledge and detailed insight This Guide introduces what you want to know about Ideal gas law A quick look inside of some of the subjects covered Atmospheric thermodynamics Overview Thermodynamic instruments Thermodynamic meters Glossary of engineering I Idealization Limits on use Perfect gas Stoichiometry Water vapor Water vapor and dry air density calculations at 0 C Equipartition theorem Perfection Physics and chemistry Glossary of chemistry terms U Fusion energy 1960s Timeline of low temperature technology 19th century Gas Avogadro's law Hot air balloon List of multiple discoveries 17th century Amount of substance Equation of state Overview Explosive Volume of products of explosion Aerodynamics Conservation laws Van der Waals equation Validity Equipartition of energy Gas Physical characteristics Gas meter Flow measurement calculations Mass flow sensor Chamber pressure Importance in Firearm Maintenance Weather forecasting How models create forecasts Timeline of hydrogen technologies 1800s Pressure Pressure of an ideal gas Compressible fluid One Dimensional Flow Diffusion Elementary theory of diffusion coefficient in gases Water vapour Water vapor and Density of airdry air density calculations at 0 C Ideal gas law Numerical weather prediction Computation Gay Lussac's law Pressure temperature law Hydrostatic equilibrium Astrophysics History of thermodynamics Birth of thermodynamics as science and much more **The Wall**

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*Ideal Gas Law, Enthalpy, Heat Capacity, Heats of Solution and Mixing* Eric H Snider, 1984-01-01      **The Ideal Gas Law Handbook - Everything You Need to Know about Ideal Gas Law** Patrick Hurley, 2016-04-29 This book is your ultimate Ideal gas law resource. Here you will find the most up-to-date information, facts, quotes, and much more. In easy-to-read chapters with extensive references and links to get you to know all there is to know about Ideal gas law's whole picture right away. Get countless Ideal gas law facts right at your fingertips with this essential resource. The Ideal gas law Handbook is the single and largest Ideal gas law reference book. This compendium of information is the authoritative source for all your entertainment, reference, and learning needs. It will be your go-to source for any Ideal gas law questions. A mind-tickling encyclopedia on Ideal gas law, a treat in its entirety, and an oasis of learning about what you don't yet know but are glad you found. The Ideal gas law Handbook will answer all of your needs and much more.

Ideal Gases Lifeliqe, 2019 This lesson plan covers the ideal gas law and the different values for the ideal gas constant, how to make various calculations using the ideal gas law, and explains the conditions under which real gases are most or least ideal.

The Gas Laws Malcolm Stubbs, Coventry University, 1995      **The Impact of High School Students' Difficulties with Operational Definitions on Understanding the Ideal Gas Law** Victor Andres Gonzalez, 2004      **The Ideal Gas**, 2015 Delve into the deep link between thermodynamics, which looks at heat on the macroscopic scale, and statistical mechanics, which views it on the molecular level. Your starting point is the ideal gas law, which approximates the behavior of many gases, showing how temperature, pressure, and volume are connected by a simple formula.

On the Definition of the Ideal Gas Edgar Buckingham, 1911      **Williams & Meyers Oil and Gas Law** Patrick H. Martin, Bruce M. Kramer, 2020      **Oil and Gas Law**, 1993

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