

TITRATIONS



Calculate the concentration of the sulfuric acid in mol/dm³

Use only the student's concordant results

mean titre = $\frac{22.10 + 22.15 + 22.15}{3} = 22.13 \text{ cm}^3$ ✓

$$n(\text{NaOH}) = 0.105 \times (22.13/1000) = 0.002324 \text{ mol} \quad \checkmark$$

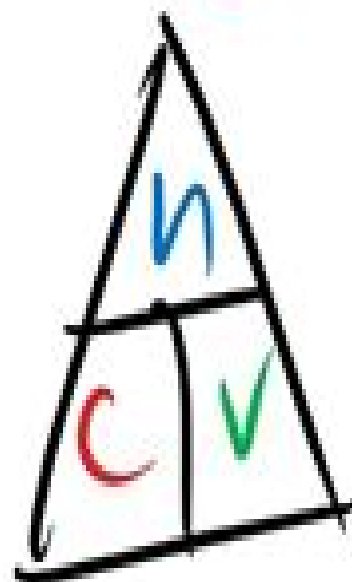
$$n(\text{H}_2\text{SO}_4) = 0.002324 / 2 = 0.001162 \text{ mol} \quad \checkmark$$

$$c(\text{H}_2\text{SO}_4) = \frac{0.001162}{(25/1000)} = 0.0465 \text{ mol/dm}^3 \quad \checkmark$$

$$n = m/M_r$$

$$n = c \times v$$

$$c = n/v$$



GCSE CHEMISTRY

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Richard Parsons

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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