



PREPARATORY EXAMINATION VOORBEREIDENDE EKSAMEN

2024

MARKING GUIDELINES NASIENRIGLYNE

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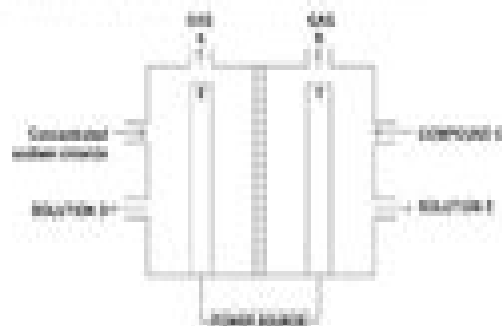
PHYSICAL SCIENCES: CHEMISTRY
FISIESE WETenskappe: CHEMIE

(PAPER/VRAESTEL 2)

15 pages/15 vaaie

QUESTION 9 (Start on a new page)

Electrolysis is generally used in industry to produce chemicals through the decomposition of compounds. The simplified diagram below represents an electrolytic cell used in the electrolysis of a concentrated sodium chloride solution.



- 9.1 Define the term *anode* in terms of oxidation or reduction. (1)

7/10

PHYSICAL SCIENCES: CHEMISTRY
PAPER 2 TIME: 2h

- 9.2 Which electrode, X or Y, is connected to the positive terminal of the power source? Give a reason for the answer. (2)
- 9.3 Write down the NAME or CHEMICAL FORMULA of
- 9.3.1 Gas A. (1)
- 9.3.2 Compound C. (1)
- 9.4 Write down the equation for the half-reaction that takes place at the cathode of the cell. (2)
- 9.5 Refer to the relative strength of the oxidising agents to explain the answer to 9.4. (2)

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1. Understanding the eBook Gauteng Prep Exam Chemistry 2014 Memo
 - The Rise of Digital Reading Gauteng Prep Exam Chemistry 2014 Memo
 - Advantages of eBooks Over Traditional Books
2. Identifying Gauteng Prep Exam Chemistry 2014 Memo
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gauteng Prep Exam Chemistry 2014 Memo
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gauteng Prep Exam Chemistry 2014 Memo
 - Personalized Recommendations
 - Gauteng Prep Exam Chemistry 2014 Memo User Reviews and Ratings
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 - Gauteng Prep Exam Chemistry 2014 Memo Budget-Friendly Options
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- ePub, PDF, MOBI, and More
 - Gauteng Prep Exam Chemistry 2014 Memo Compatibility with Devices
 - Gauteng Prep Exam Chemistry 2014 Memo Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Gauteng Prep Exam Chemistry 2014 Memo
 - Highlighting and Note-Taking Gauteng Prep Exam Chemistry 2014 Memo
 - Interactive Elements Gauteng Prep Exam Chemistry 2014 Memo
8. Staying Engaged with Gauteng Prep Exam Chemistry 2014 Memo
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gauteng Prep Exam Chemistry 2014 Memo
9. Balancing eBooks and Physical Books Gauteng Prep Exam Chemistry 2014 Memo
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gauteng Prep Exam Chemistry 2014 Memo
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gauteng Prep Exam Chemistry 2014 Memo
- Setting Reading Goals Gauteng Prep Exam Chemistry 2014 Memo
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gauteng Prep Exam Chemistry 2014 Memo
- Fact-Checking eBook Content of Gauteng Prep Exam Chemistry 2014 Memo
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

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- Interactive and Gamified eBooks

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