

AMINO ACID WORKSHEET

Name:

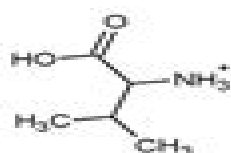
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- Using the amino acid chart on the back of this worksheet, divide the amino acids into the following groups: hydrophobic, hydrogen bonding, acidic, basic, amide, small, able to form sulfide bonds. List the symbol for the amino acids in the space provided. Use the third row of the chart to indicate the role the amino acids in this group will play in the structure of the protein or other special features. Not all amino acids will be used.

hydrophobic	hydrogen bonding	Acidic or amides	basic	disulfide bond	imino acid
Gly, Ala, Val, Leu, Ile, Phe, Met, Trp	Gln, Asn, Ser, Thr, Tyr, Pro	Asp, Glu, *Cys <i>Asn, Gln</i>	His, Lys, Arg	Cys	Pro
Hydrophobic interactions contribute to tertiary and quaternary structure. Hydrophobic amino acids are mostly in the interior of the protein	Hydrophilic interactions contribute to the tertiary and quaternary structure; found on the inside and outside of the proteins	When side chains become charged due to pH, can lead to ionic attraction; found on outside of protein; tertiary and quaternary structure		Disulfide bonds can effect primary, tertiary and quaternary structure	Forms "kinks" in secondary structure of the protein

- Draw each amino acid below at pH = 1.0. Then determine its charge at this pH.

valine



charge = +1

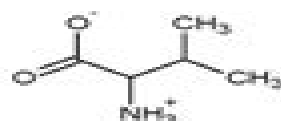
aspartate



charge = +1

- Draw each amino acid below at pH = 7.4 (pH of blood). Then determine its charge at this pH.

Valine



Charge = 0

aspartate



charge = -1

Lab Amino Acid Answer Key

**Dr. Prashant D. Aragade, Dr. N. B.
Ghiware ,Mr. Sushant Sudhir
Pande, Dr. Seemadevi Suresh Kadam**

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Laboratory Manual of Analytical Methods of Protein Chemistry P. Alexander, H. P. Lundgren, 2014-05-09 A Laboratory Manual of Analytical Methods of Protein Chemistry Volume 4 provides information pertinent to the fundamental aspects of protein chemistry This book discusses the simple and accurate methods of estimating specific proteins Organized into six chapters this volume begins with an overview of the composition of acids and experimental conditions for the acid hydrolysis of proteins This text then examines the advantages of high voltage electrophoresis for amino acid analysis which are paralleled by equal advantages in the peptide separation field Other chapters consider the simple technique of estimating specific proteins which is one of several based on the phenomenon of antigen antibody precipitation in gels This book discusses as well the summations of analyses in weight percentages of the various residues and of the nitrogen of each constituent The final chapter deals with the electrical properties of molecules This book is a valuable resource for physicists

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 Aragade, Dr. N. B. Ghiware, Mr. Sushant Sudhir Pande, Dr. Seemadevi Suresh Kadam, 2024-02-16 We are pleased to put forth
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1. Understanding the eBook Lab Amino Acid Answer Key
 - The Rise of Digital Reading Lab Amino Acid Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Lab Amino Acid Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Lab Amino Acid Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Lab Amino Acid Answer Key
 - Personalized Recommendations
 - Lab Amino Acid Answer Key User Reviews and Ratings
 - Lab Amino Acid Answer Key and Bestseller Lists
5. Accessing Lab Amino Acid Answer Key Free and Paid eBooks
 - Lab Amino Acid Answer Key Public Domain eBooks

- Lab Amino Acid Answer Key eBook Subscription Services
- Lab Amino Acid Answer Key Budget-Friendly Options
- 6. Navigating Lab Amino Acid Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Lab Amino Acid Answer Key Compatibility with Devices
 - Lab Amino Acid Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Lab Amino Acid Answer Key
 - Highlighting and Note-Taking Lab Amino Acid Answer Key
 - Interactive Elements Lab Amino Acid Answer Key
- 8. Staying Engaged with Lab Amino Acid Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Lab Amino Acid Answer Key
- 9. Balancing eBooks and Physical Books Lab Amino Acid Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Lab Amino Acid Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Lab Amino Acid Answer Key
 - Setting Reading Goals Lab Amino Acid Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Lab Amino Acid Answer Key
 - Fact-Checking eBook Content of Lab Amino Acid Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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