

Experiment 6B: Calculations with a Chemical Reaction

Objectives:

1. To understand the theory of stoichiometry, including the concept of molar mass and limiting reagents, for stoichiometric calculations.
2. To calculate the limiting reagent, identifying the limiting reagent present in the mixture.
3. To calculate stoichiometrically the maximum yield of a reaction.
4. To calculate the theoretical maximum stoichiometric reagents that could be produced.
5. To compare the theoretical amount of the actual amount of reagents stoichiometrically.

Materials: Please refer to Experiment 6B page 95 of the Practical Chemistry Laboratory

Equipment:

Procedure:

1. Prepare very carefully experiment with the people's equipment.
2. Prepare a solution of 100 ml of water with a concentration of 100 g/l.
3. Calculate 10 ml of concentrated solution which is produced by 100 g/l.
4. Dilute 100 ml of the solution to 100 ml. Then the produced 100 ml of solution.
5. Use the 100 ml of the solution to produce the solution.
6. Prepare 100 ml of concentrated solution and use the solution.
7. Use the 100 ml of the solution to produce the solution. Then the solution.
8. Use the 100 ml of the solution to produce the solution. Then the solution.
9. Prepare the 100 ml of the solution to produce the solution.
10. Prepare the 100 ml of the solution to produce the solution.

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