

Experiment #12: Acid-Base Titration

NAME: _____

Section: _____

Date: _____

DATA & LAB REPORT:

Part A: Determining Molarity of NaOH Solution

*Show ALL Calculation Setups

	Trial 1	Trial 2	Trial 3
Mass of KHP	0.5858 g	0.9510g	
Final Buret Reading	31.10mL	44.10 mL	
Initial Buret Reading	0.73 mL	0.00 mL	
Volume of NaOH Used	30.37 mL	44.10 mL	

1. Calculate moles of KHP used in each trial.

Moles of KHP

Trial 1	Trial 2	Trial 3
_____	_____	_____

2. How many moles of NaOH was used in each trial?

Moles of NaOH

Trial 1	Trial 2	Trial 3
_____	_____	_____

Lab 39 Acid Base Titration Answers

AW Chickering



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