

F324-unofficial mark scheme (in progress)

Marks accounted for: 60/60

- Give the systematic name of the alcohol that is formed by this hydrolysis. (1)
 - Propan(e)-1,2,3-triol
- What are the other monomers that make up this polymer? (3)
(Hydrolysed by hot NaOH)
 - Two different carboxylic acids with $\text{COO}^-(\text{Na}^+)$ - The monomers have COOH as it's not referring to the hydrolysed products but the monomer units of the polymer I'm pretty sure that the question said: "What are the other products of this hydrolysis" or something along those lines
 - One salt had a $\text{C}=\text{C}$ double bond
- Why are people who drink a lot of goat's milk likely to suffer from CHD? (2)
 - Fatty acids build up in the arteries and lead to CHD?
 - Has trans isomer due to $\text{C}=\text{C}$ double bond fatty acid linked to LDL's and hence Atherosclerosis
 - Trans-fatty acid in triglyceride of milk increases 'bad' cholesterol levels in blood, resulting in increased risk of coronary heart disease and stroke (straight from the spec)
 - trans fatty acids can pack more tightly and hence have higher melting points than cis fatty acids. \rightarrow so more likely to deposit on artery walls
 - The other two molecules were fully saturated fatty acids which would also lead to the formation of atheroma
- State how 4-aminophenol can behave as a base. (1)
- The lone pair on the Nitrogen atom in the amine group can accept a proton, forming a dative covalent bond with it to give NH_3^+ .
- $\text{NH}_2 + \text{H}^+ \rightarrow \text{NH}_3^+$
- 4-Aminophenol is produced by the reduction of 4-nitrophenol. Write an equation to show the production of 4-aminophenol from 4-nitrophenol. Use $[\text{H}]$ to represent the reducing agent. (1)
- $4\text{-Nitrophenol} + 6[\text{H}] = 4\text{-Aminophenol} + 2\text{H}_2\text{O}$
- 4-nitrophenol can be produced from 4-bromophenol. (4)
 - Complete the mechanism for this reaction.
 - Use $^+\text{NO}_2$ as the electrophile. Include any intermediate and the products.
 - In the mechanism for this reaction, NO_2 substitutes for Br on the ring.

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Xiaolong Qi



Gcses June 2013 Unofficial Mark Scheme Chem:

Enjoying the Tune of Phrase: An Emotional Symphony within **Gcses June 2013 Unofficial Mark Scheme Chem**

In a world used by monitors and the ceaseless chatter of immediate communication, the melodic beauty and psychological symphony produced by the written word usually disappear into the background, eclipsed by the persistent sound and disruptions that permeate our lives. But, nestled within the pages of **Gcses June 2013 Unofficial Mark Scheme Chem** an enchanting fictional treasure full of fresh thoughts, lies an immersive symphony waiting to be embraced. Constructed by a wonderful musician of language, that fascinating masterpiece conducts readers on a mental journey, skillfully unraveling the hidden tunes and profound impact resonating within each cautiously crafted phrase. Within the depths of the emotional evaluation, we shall explore the book is central harmonies, analyze their enthralling writing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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Table of Contents Gcses June 2013 Unofficial Mark Scheme Chem

1. Understanding the eBook Gcses June 2013 Unofficial Mark Scheme Chem
 - The Rise of Digital Reading Gcses June 2013 Unofficial Mark Scheme Chem
 - Advantages of eBooks Over Traditional Books
2. Identifying Gcses June 2013 Unofficial Mark Scheme Chem
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gcses June 2013 Unofficial Mark Scheme Chem
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gcses June 2013 Unofficial Mark Scheme Chem
 - Personalized Recommendations

- Gcses June 2013 Unofficial Mark Scheme Chem User Reviews and Ratings
- Gcses June 2013 Unofficial Mark Scheme Chem and Bestseller Lists
- 5. Accessing Gcses June 2013 Unofficial Mark Scheme Chem Free and Paid eBooks
 - Gcses June 2013 Unofficial Mark Scheme Chem Public Domain eBooks
 - Gcses June 2013 Unofficial Mark Scheme Chem eBook Subscription Services
 - Gcses June 2013 Unofficial Mark Scheme Chem Budget-Friendly Options
- 6. Navigating Gcses June 2013 Unofficial Mark Scheme Chem eBook Formats
 - ePub, PDF, MOBI, and More
 - Gcses June 2013 Unofficial Mark Scheme Chem Compatibility with Devices
 - Gcses June 2013 Unofficial Mark Scheme Chem Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gcses June 2013 Unofficial Mark Scheme Chem
 - Highlighting and Note-Taking Gcses June 2013 Unofficial Mark Scheme Chem
 - Interactive Elements Gcses June 2013 Unofficial Mark Scheme Chem
- 8. Staying Engaged with Gcses June 2013 Unofficial Mark Scheme Chem
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gcses June 2013 Unofficial Mark Scheme Chem
- 9. Balancing eBooks and Physical Books Gcses June 2013 Unofficial Mark Scheme Chem
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gcses June 2013 Unofficial Mark Scheme Chem
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gcses June 2013 Unofficial Mark Scheme Chem
 - Setting Reading Goals Gcses June 2013 Unofficial Mark Scheme Chem
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gcses June 2013 Unofficial Mark Scheme Chem
 - Fact-Checking eBook Content of Gcses June 2013 Unofficial Mark Scheme Chem

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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