

(iii) Suggest why the mean bond enthalpy of a C=C bond is less than twice the mean bond enthalpy of a C—C bond.

(1)

(iv) Use the mean bond enthalpy data in the table, and the equation given below, to calculate a value for the standard enthalpy change of combustion of propene.

(3)

Bond	Mean bond enthalpy / kJ mol ⁻¹
C=C	612
C—C	347
C—H	413
O=O	498
C=O	805
O—H	464



Answer = _____ kJ mol⁻¹



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Table of Contents Gce Edexcel Chemistry 2015 January Papers

1. Understanding the eBook Gce Edexcel Chemistry 2015 January Papers
 - The Rise of Digital Reading Gce Edexcel Chemistry 2015 January Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying Gce Edexcel Chemistry 2015 January Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gce Edexcel Chemistry 2015 January Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gce Edexcel Chemistry 2015 January Papers

- Personalized Recommendations
- Gce Edexcel Chemistry 2015 January Papers User Reviews and Ratings
- Gce Edexcel Chemistry 2015 January Papers and Bestseller Lists
- 5. Accessing Gce Edexcel Chemistry 2015 January Papers Free and Paid eBooks
 - Gce Edexcel Chemistry 2015 January Papers Public Domain eBooks
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 - Gce Edexcel Chemistry 2015 January Papers Budget-Friendly Options
- 6. Navigating Gce Edexcel Chemistry 2015 January Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - Gce Edexcel Chemistry 2015 January Papers Compatibility with Devices
 - Gce Edexcel Chemistry 2015 January Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gce Edexcel Chemistry 2015 January Papers
 - Highlighting and Note-Taking Gce Edexcel Chemistry 2015 January Papers
 - Interactive Elements Gce Edexcel Chemistry 2015 January Papers
- 8. Staying Engaged with Gce Edexcel Chemistry 2015 January Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gce Edexcel Chemistry 2015 January Papers
- 9. Balancing eBooks and Physical Books Gce Edexcel Chemistry 2015 January Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gce Edexcel Chemistry 2015 January Papers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gce Edexcel Chemistry 2015 January Papers
 - Setting Reading Goals Gce Edexcel Chemistry 2015 January Papers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gce Edexcel Chemistry 2015 January Papers

- Fact-Checking eBook Content of Gce Edexcel Chemistry 2015 January Papers
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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