# Mark Scheme

#### QU.

Question	Acceptable Answer	Additional Guidance	Mark
(a) (i)	$CaCO_2(s) + 2HCl(eq) \rightarrow CaCl_2(eq) + H_2O(t) + CO_2(g)$	Accept $CaCO_5(s) + 2H^*(aq) \rightarrow Ca^{2^*}(aq) + H_5O(l) + CO_5(g)$	(5)
	Balanced equation (1)		
	State symbols (1)	2nd mark dependent on first or near miss.	
		Reject HyCO <sub>2</sub> (aq) in equation, but allow state symbol mark if otherwise correct.	

Question	Acceptable Answer	Additional Guidance	Mark
(a) (iii)	Finds molar mass of calcium carbonate	Example of calculation  Mr of calculation carbonate = 40.1 + 12 + (16 x 3) = 100.1 (g mof <sup>4</sup> )  Allow = 40 + 12 + (16 x 3) = 100 (g mof <sup>4</sup> )  Accept answer with no working	(1)

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