Model 2 – Prefixes and Suffixes

| Prefix | Numerical Value |
|--------|-----------------|
| mhomo- | 1 |
| di- | 2 |
| eri- | 3 4 |
| octra- | 4 |
| penta- | 5 |
| hexa- | 6 |
| hepsa- | 7 |
| octa- | 8 |
| morka- | 9 |
| deca- | 10 |

| Molecular Formula | Name of Compound |
|-------------------|--------------------------|
| BCI, | Boron trichloride |
| SF. | Sulfur hexafluoride |
| IF. | Iodine heptafluoride |
| NI, | Nitrogen triiodide |
| N,O, | Dinitrogen tetroxide |
| CI,O | Dichlorine monoxide |
| P.O.,. | Tetraphosphorus decoxide |
| B.H. | Pentaboron nonahydride |
| Br,O. | Tribromine octoxide |
| CIF | Chlorine monofluoride |

- 5. Examine the prefixes in Model 2. Hill in the numerical value that corresponds to each prefix.
- 6. What suffix (ending) do all the compound names in Model 2 have in common? They all end in "ide"



- 7. Carefully examine the names of the compounds in Model 2. When is a prefix NOT used in from of the name of an element of the name of an element of the first element is one. (Examples: IFn-locline high flow the Wig-nitrogen triid ide
 - 8. Consider the compound NO.
 - a. Which element, nitrogen or oxygen, would require a prefix in the molecule name? Explain

Oxygen would require a prefix of "mono" because it is the second element and there is one oxygen Nitrogen does not require a prefix because it is the first element and there is only a Name the molecule NO. nitroser.

nitrogen monoxide (Notice, the "o" of minous dige 9. Find two compounds in Model 2 that contain a subscript of "4" in their molecular formula.

- - N204 dinitrogen tetroxide II P4010-Tetrophosphorus decoxide a. List the formulas and names for the two compounds.
 - 6. What is different about the spelling of the prefix meaning "four" in these two names? The "a" has been dropped from the end of the profix tetra.

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